

BASE SOLUBILITY AND MARGINAL SEALING IN AMALGAM RESTORED TEETH

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Marginal leakage between freshly packed dental amalgam restorations and the cavity wall decreases with time due to the development of a marginal seal which fills the space at the tooth-amalgam interface¹. It is believed that this seal develops in part from dental materials used in cavity preparation such as the base, varnish and amalgam². The role which the base plays in sealing is obscure. It could be that the disintegration which this material undergoes in the mouth physically blocks the marginal gap. This *in vitro* study was undertaken to investigate the effect of base solubility on marginal seal development and composition with time.

Occlusal cavities were cut in 192 caries-free extracted human teeth. Cavities were lined with one of four bases, which were either predominantly of calcium hydroxide or zinc oxide. The zinc oxide base is hard and insoluble when mixed with eugenol (ZOE) or soluble when mixed with distilled water (ZO). Similarly the calcium hydroxide containing base is insoluble as Dycal^R (DY) or soluble when calcium hydroxide is mixed with distilled water (CAL). A varnish was applied to half of the cavities followed by restoration with a low copper amalgam (Cu(low)) to produce 8 treatment combinations. The teeth were stored in 1% NaCl and thymol for 7 days, 3 months and 1 year at 20°C where after a standard fluorescent dye marginal leakage test was performed. These results were subjected to a SAS Linear Logistical Statistical Analysis (CATMOD). Two specimens of each 3 month and 1 year group were fractured to expose the restoration and cavity surfaces which were covered with marginal seal material. Quantitative area analysis was performed on both these surfaces using a peak-background routine to determine the elemental composition of the seal material. Viewing of the specimen took place in a JSM 840 scanning electron microscope equipped with a LINK AN10000 energy dispersive x-ray analyser. The analysis conditions were set at an operating voltage of 20kV, a count rate of 2800 counts/sec and a live counting time of 100 sec. In between each two analyses a reference spectrum was recorded from the cobalt standard in order to calibrate the energy scale. Peak to background ratios were obtained for each element and percentage elemental composition was then calculated for each analysis. The data was statistically analysed with one way ANOVA and Tukey's test using the General Linear Models Procedure. The critical level of significance was chosen at $p \leq 0.05$ for all statistical procedures. Similar analyses were carried out on all the dental materials used as well as on the cut tooth cavity surface to obtain reference spectra to serve as comparative standards.

