

Abstract

The new chelating bis-phosphine ligands 1,2-bis(butylphenylphosphino)ethane (**bppe**) and *cis*-1,2-bis(butylphenylphosphino)ethylene (**bppey**) have been synthesised by (i) a one-pot synthesis from Ph₂PBu and Li/dihalo alkyl, (ii) the reaction of 1,2-bis(diphenylphosphino)ethane (**dppe**) or *cis*-1,2-bis(diphenylphosphino)ethylene (**dppey**) and Li/*n*-BuCl and, (iii) in the case of **bppe** by sequential synthesis (all intermediates were isolated) from Ph₂PBu *via* PhBuPLi and Ph₂PH. **bppe** and **bppey** as well as the new lithium phosphide [(TMEDA)•LiPPh(Bu)]₂ (**17**) were fully characterised by multinuclear NMR spectroscopy, mass spectrometry and, in the case of [(TMEDA)•LiPPh(Bu)]₂ (**17**), by X-ray crystallography.

Reaction of the bis-phosphines **bppe** and **bppey** with suitable metal precursors yielded the corresponding metal complexes: [PdCl₂(bppe)] (**18**), [Pd(bppe)₂](ClO₄)₂ (**20**), [(AuCl)₂(bppe)] (**21a**), [(AuCl)₂(bppey)] (**21b**), [Au(bppe)₂]Cl (**22a**), [Au(bppey)₂]Cl (**22b**), [(AgNO₃)₂(bppe)] (**23**) and [Au(bppe)₂]ClO₄ (**24**) in moderate to good yields. All were characterised by multinuclear NMR spectroscopy and mass spectrometry, while **18** and **20** were further characterised by X-ray crystallography.

Preliminary stability tests showed, that of all new metal complexes only **18**, **20** and **22a** were adequately stable to justify further tests for anti-tumour activity. The cationic complexes **20** and **22a** showed activity against HeLa cells while the neutral complex **18** was not active. A comparison with the previously investigated analogous **dppe** and **dppey** complexes revealed that **20** and **22a** were found to be less active as a result of the replacement of a Ph groups with butyl groups in the phosphine ligand.