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Domonique Basson; Sandra Van Vuuren; Ida Risenga

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AUTHORS:
Domoni{lue C. Bisson' G>
Sandy var Vuuren' G>
Ida M Risenga' G

AFFILIATIONS:
'Schoolof Animal, Plantand Environmental Scie11ces.University of the Witwatersrand, Johannesburg, South Africa
[-1lepartment of Pharmacyand Pharmacology,Universi!yof the Witwateri.rand, Johannesburg, South Alrica

CORRESPONDENCE TO:
Ida Risenga

EMAIL:
Ida.Rise11ga@wits.ac.za

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The effect of elevated carbon dioxide on the medicinal properties of *Portulacaria afra*

There is a global concern thal rising atmospheric $C0_1$ concentrations may impact the medicinal or nutritional properties of medicinal plants. *Portulacaria afra* is a South African medicinal plant that is used by traditional healers to treat various skin conditions. The aim of this study was to determine whether elevated CO_2 concentrations would affect the medicinal properties of the leaves of *P. alra.* This was achieved by comparing the phytochemical presence, antioxidant and antimicrobial activity of the leaves of *P. afra* which were exposed to ambient (420 ppm) and elevated (600 ppm) CO_2concentrations of plants grown in greenhouse conditions. The results revealed that leaf samples that were exposed to elevated CO_2 concentrations exhibited a significant increase in flavonoid presence, compared to the control group. The antioxidant activity of the leaves of *P. afra* (OPPH activity) remained mostly unchanged in the samples that were exposed to elevated CO_2 concentrations. The antimicrobial activity efficacy against *Cutibacterium acnes* increased with increasing global atmospheric $C0_1$ concentration. These findings suggest that *P.afra* is a resilient medicinal plant and that its leaves may continue to provide relief against certain ailments, despite rising atmospheric CO_2concentrations.

Significance:

- *Portulacaria alra* is a South African medicinal'yimportant species that shows great resilience against elevated CO_2concentrations.

  It is important to anticipate how changing environmental factors, such as rising CO_2concentrations, may attect natural resources.
- The phytochemical profile and antioxidant and antimicrobial activities of the various plant parts either remained the same or increased after exposure to an e evatedCO_2concentration of 600 ppm.

Introduction

Prior to the industrialisation era, the atmospheric carbon dioxide (COz) concentration remained relatively stable at 280±10 parts per million (ppm).[1] In the year 2022, the atmospheric CO_2 concentration reached an average of 417.2 ppm, representing a more than 50% increase. Currently, the energy sector remains heavily reliant on fossil fuels, and as a result, this value is predicted to continue rising at an unprecedented rate, potentially reaching 600 ppm by 2050.[2]

Carbon dioxide is an essential resource for the growth, development and overall survival of plants, as it forms a major component ol photosynthesis - which produces sugars, carbohydrates and other organic molecules necessary for plant functiorr. As CO_2 is a limiting factor in photosynthesis, the rise in atmospheric CO_2may result in an increase in the photosynthetic pathway. Previous studies have shown that elevated CO_2 concentrations can lead to an alteration of primary and secondary metabolism,resulting in an increase of up to 44% in plant biomass.[4] However, the alteration to secondary metabolites may result in an unequal distribution of chemicals throughout the plant. The increase in CO_2concentration results in an accumulation of carbon-based phytochemicals, such as phenolics, phlobatannins and flavonoids.[3]

This increase in carbon-based molecu lescan lead to a decrease in thepartitioning and allocation of other molecules to plant organs such as leaves.[l] Scientists warn that the alteration in the chemical composition of the plant may affect the potency of the medicinal properties of the plant.$

In a previous study, *Asclepias curassav1ca* was subjected to elevated CO_2concentrations and ultimately exhibited a decrease in medicinal properties. However,this is not a consistent trend in the literature.[6] The ettect of elevated $C0_1$ concentrations varies with species exhibiting a negative impact, a positive impact or no impact at all.[7] Thus, more studies need to be conducted on the effect of rising CO_2concentrations on various medicinal plant species.

Portulacana atra is a medicinal plant species endemic to South Africa and is also found in neighbouring countries such as Eswalini and Mozambique.• Despite being primari'y known for its carbon sequestration capabilities,• ethnobotanical interviews conducted 111 2013 revealed that *P. atra* is used traditionally to treat various skin conditions." The small succulent leaves are predominately used topically to treat conditions such as rashes, ringworm and acne [11]

Given that a large population still relies on medicina plants as a primary source of health care, it is imperative to determine the ettect that rising CO_2concentrations may have on these resources. In this study, we aimed to assess the ettects of an elevated atmospheric CO_2concentration on the physiological properties, phytochemical profile and biological activity of the leaves of *P.afra.*Inorder to accomplish this, *P.afra* plants were exposed to both ambient CO_2concentration (420 ppm) and elevated CO_2 concentration (600 ppm) to determine whether there was a significant difference in the medicinal properties of the plant. Indicators of medicinal properties used in this study include the presence of 10 phytochemical groups, antioxidant activity against DPPH and HA, and antimicrobial properties against pathogens commonly associated with skin infections.

Materials and methods

Plant material

Portulacaria afra cuttings that exhibtted no signs of wHling or disease were collected from the University of the Witwatersrand (26.1929° S, 28.0305" E) in December 2021 (South African summer time). The plants were identified by a botanist (IMR). A voucher specimen was deposited in the institute's herbarium (IMROO1). The 90 *P. afra* plants were maintained in the Oppenheimer Life Sciences greenhouse, at the University of the Witwatersrand.

Treatment

Of the 90 plants tiat weremaintained in the greenhouse, 30 were harvested and directiy subjected to analysis as control samples. The remaining 60 plants were transferred to a Conviron® climate simulator to expose the plants to varying CO_2concentrations. Of these, 30 were exposed to a CO_2 concentration of 420 ppm and the remaining 30 plants were exposed to a CO_2concentration of 600 ppm.The temperature, light and relative humidity remained ambient and constant between the treatments. The plants were watered every second or third day, wilh approximately 200 ml water.

Each month, for a period of 3 months, 1O plants were harvested from each chamber and subjected to analysis.

Chlorophyll content

The fresh leaves (3 g) were mixed with 10 ml of 80% methanol and incubated in the dark for 24 h. The supernatant was then collected, and absorbance was measured using a spectrophotometer (Genesys 1OS UV-VIS) at 645 nm and 663 nm to obtain the absorbance values. The chlorophyll content was calculated using Equation 1:

$$20.21\ xA_{645} + 8,02\ xA \qquad \text{Equation 1}$$

This method followed the procedurn ouUined by Liu et a1.[1]

Extraction

The crude plant extract used to determine the phytochemical presence and antioxidant activity was prepared using distiled water at 40 •C,80% methanol, dichloromethane or n-hexane. These solvents were chosen due to the varying polarities between the solvents. The fresh leaves were placed in an air dryer at 40 •C for the duration of a week, or until sufficienUy dry. Once sufficiently dried, the plant material was placed into a mechanical grinder to create powdered plant material. A mixture of 3 g of the powdered plant material and 30 ml of the respective solvent was placed on a mechanical shaker for 48 h. The hot water extracts were placed on a hot plate set at 40 •C and stirred with a magnetic stirrer. The extracts were filtered through a Whatman No. 2 filter paper. The plant extracts were stored at 4 °C prior to phytochemical and antioxidant analysis. The crude plant extracts used to determine the antimicrobial activity of *P. alra* were created by mixing powdered plant with dichloromethane:methanOl at a ratio of 1:1.[13]

Preliminary phytochemlcal screening

A qualitative analysis was conducted to determine the presence or absence of 1O phytochemical groups in the leaves of ***P. alra*** using standard test methods.[14,15] The 1O phytochemical groups considered were saponins, phenolics, llavonoids, glycosides, tannins terpenoids, steroids, coumarins, phlobataMins and volatile oils.

Total llavonoid content

An aluminium chloride colourinetric assay was used to determine the total flavonoid content in the leaves, stems and roots of *P.afra."*

In a test tube, 3 ml of the crude plant extracts were mixed with 4 ml of 5% sodium nitrate. The mixt..-e was incubated for 5 min. Following the incubation period, 3 ml of 10% aluminium chloride was added to the sample and incubated for an additional 6 min. A sodium hydroxide solution with a volume of 2 ml was then added to the sample. The sample was topped with O.7 ml of distilled water to create a mixture of 10 ml. The test tubes were incubated at ambient temperature for 1 h. The absorbance of each mixture was measured at 510 nm using the Genesys 1OS UV-VIS spectrophotometer.

The test was conducted in triplicate. A caHbralion curve was created using a quercelin standard. The total flavonoid content was calculated from the calibration curve using Equation 2:

$$Y = 0.2388x - \mathbf{O}.0019 \qquad \text{Equation 2}$$

Antioxidant activity

DPPH scavenging activity

The stable radical OPPH was used lo determine the scavenging activity and hence the antioxidant activity of the leaves of *P. alra*.[11] Varying volumes (10-50 µL) of the crude plant extract were placed in capped test tubes and mixed with 700 µL of the OPPH work solution. In addition, 80% methanol was used to create a mixture of 1 ml. The samples were incubated in the dark, at ambient temperature for 45 min. Following the incubation period, the mixtures were placed into cuvettes that were placed within a spectrophotometer (Genesys 1 OS UV-VIS). A cuvette containing 1 ml of 80% methanOI was used as the 'blank' to zero the spectrophotometer. The absorbance was measured at 517 nm against a blank. The test was then done in triplicate and thepercentage of inhibition was calculated using Equation 3:

$$\%\ \text{inhibition of DPPH radical} = ((A_{llf} - A.,)/\mathbf{A.,}\} \times 100 \qquad \text{Equation 3}$$

where *Allf* is the absorbance of the control and *A.,*is the absorbance of the sample. The concentrations were then plotted against percentage inhibition values. Microsoft Excel was used to perform a linear regression to obtain the ICic. R-studio was used to perform a repeated measures ANOVA. All statistical tests were conducted at a significance level of 0.05.

H_2O_2 scavenging activity

The free radical H_2O_2 was used to determine the scavenging activity and hence the antioxidant activity of leaves of *P. alra.*[17]A 40 mM solution of H_2O_2 (hydrogen peroxide) was prepared in a phosphate buffer (pH 7.4). The crude plant extracts (10-50 µl) were placed in vials and mixed with 600 µL of the hydrogen peroxide solution. The vials were then placed in the dark and incubated for 10 min. The absorbance of the mixtures was determined using a spectrophotometer (Genesys IOS UV-VIS) at 230 nm. A cuvette containing 1 ml of the phosphate buffer without hydrogen peroxide was used as the 'blank' to zero the spectrophotometer. This test was oodertaken in triplicate. The different absorbances were recorded. The percentage of hydrogen peroxide scavenging activities was calcu1atedusing Equation 4:

$$\text{"Scavenged } (H_2O_2) = ((A..-A.)/A.,\} \times 100 \qquad \text{Equation 4}$$

where *A.,* is the absorbance of the control and *A.* is the absorbance of the samp e. The concentrations were then plotted against percentage inhibition values. Microsoft Excel was used to perform a linear regression to obtain the ICIMI. R-studio was used to perform a repeated measures ANOVA. All statistical tests were conducted at a significance level of O 05.

Antimicrobial ***assay***

Culture **preparations**

Bacteria microorganisms selected for the study were *Slaphy/<Jcaccus aureus* (ATCC 25923), *Staph-jlococcus eptdermidis* (ATCC 1228), *Cutibacterium acnes* (ATCC 29212), *Pseudomonas aeruginosa* (ATCC *27853),andKlebsiellaaerogenes* (ATCC13048). Theyeastmicrl JO(ganism was *Candida albicans* (ATCC 10231). These microorganism strains are all pathogens pertaining to skin infections and were obtained from the American Type Culture Collection from the Department of Pharmacy and Pharmacology, University of the Witwatersrand, Johannesburg.

Minimum inhibitory concentration antimicrobial assay

The twofold serial dilutionmicrodilution technique wasused to determine the lowest concentration ol the plant extracts that would inhibit the growth ol the selected microorganisms, that is, the minimum inhibitory concentration (MIC). A 96-well microtitre plate was prepared by adding 1DD µL of the respective broth into the wells. A volume of 1DD µL of the plant extract was placed into the first well (A) using an aseptic technique. A positive control (ciprolloxacin for bacteria or nystatin for the yeast), negative control (dichloromethane: methanol) and culture control (TSB or TGB) were included. The posttive control was used to detect the microbial susceptibility of the pathogens. The negative control was used lo detect whether the pathogen was reacting to the solvent. The culture control was to ensure that microbial growth did occur.

A serial dilution was conducted at concentrations of 8, 4, 2, 0.5, 0.25, 0.125 and 0.0625 mg(ml. A 0.5 McFarland turbidity standard was created by mixing the bacterial culture with the broth at a ratio of 1:100. A volume ol 100 µL of the culture was then placed in each well. The microtitre plates were sealed with a sterile adhesive film and incubated al the respective optimum conditions. The optimum conditions for bacterial growth were 37 •C for 24 h, with the exception of *C. acnes* which was incubated al 37 •C tor 96 h. *C. albicans* was incubated at 37˙C for 48 h.

Following the incubation periods, 40 µL ol p-Iodonitrotetrazolium violet solution (INT) was added to all the wells in the microtitre plate. The wells that exhibited microbial growth (shown by a purple-pink colour) were recorded, and the MIC was taken as the lowest concentration inhibiting microbial growth.

Results and discussion

Weight and chlorophyll content

The weight ol the leaves increased in the plants that were exposed to both 420 ppm and 600 ppm of CO_2 (Table 1). However, the leaves exposed to 420 ppm exhibited a weight increase ol 23.6% (p>D.05), whereas the weight of the leaves exposed to an elevated CO_2 exhibited a 65.7% increase in weight between harvest 1 and harvest3 (p<0.05). In a previous study, it was determined that Crassulacean acid metabolism (CAM) species (which include *P. afra)* exhibited an increase in plant biomass when exposed to CO_2concentrations of 650-750 ppm for 3 months.[18] The results of this study are similar to those in the literature. This confirms the general trend that increased CO_2 may have on lhe physiological properties of *P. afra.*

Chlorophyll has a wide range of nutritional and health-promoting benefits, as a result of it being rich in vitamins and minerals, as well as exhibiting antioxidant and anti-inflammatory properties.[19]The chlorophyll content of the leaves of *P. atra* remained relatively constant in both treatments, across all three harvests (p>D.05). This suggests that the plant may offer health-promoting benefits despite being exposed to elevated CO_2 concentrations.

Phytochemical profile

The control sample (greenhouse) of *P. atra* exhibited the presence ol 8 out of the 10 phytochemicals tested (Table 2). A high level of coumarins was detected incontrol plant samples. In contrast, the test for llavonoids and phlobatannins yielded an 'absent' result, indicating no presence of these phytochemicals or a presence too low to be detected by these phytochemical analyses. The plant samples that were exposed to a CO_2 concentration of 420 ppm exhibited a phytochemical profile similar to that of the control samples.

The samples exposed to 420 ppm experienced a general decrease in phytochemical presence. After harvest 1, the saponins, phenolics and coumarins exhibited an increase in presence. However, alter harvests 2 and 3, the presence was similar to or lower than the greenhouse samples. The glycosides were the only group to have a greater presence inharvests 1, 2 and 3 thanthe control. Similar to lhe greenhouse extracts, flavonoids and phlobatannins **were** recorded as absent throughout all three harvests. The samples that were exposed to 420 ppm exhibited a lower phytochemical presence than the samples exposed to 600 ppm for 5 out of the 10 phytochemicals analysed.

The plants that were exposed lo a CO_2 concentration of 600 ppm exhibited a general increase in phytochemical presence compared to the control and samples exposed to 420 ppm. Terpenoids were the only phytochemical group that decreased from present to absent in harvests 2 and 3. Other phytochemicals, such as saponins, flavonoids, glycosides, tannins, coumarins and volatile oils, exhibited a stronger phytochemical presence.

The most notable increase was in the flavonoid group, which was documented as absent in the greenhouse and 420·ppm samples, but present and moderately present in the methanolic and hot water samples that were exposed to 600 ppm. Upon further investigation, the quantification of flavonoids revealed lhat flavonoids were not absent in the greenhouse and 420-ppm extracts, just present in low quantities (Table 3).

The leaf samples that were exposed to elevated CO_2 concentrations exhibited a significant increase in flavonoid content in the methanolic, hexanic and hot water extracts (p<0.05). The highest flavonoid content was recorded in the methanolic leaf extracts that had been exposed to 600 ppm (19.27±0.10 mg OE/g), All four 420-ppm treatment extracts exhibited an inttial decrease in flavonoid content in harvest 1. Similar to the phytochemical presence of saponins, phenolics and coumarins, harvests 2 and 3 produced llavonoid contents similar to that of the greenhouse samples. This change alter harvest 1 may allude to an adaptation period for the plant. The flavonoid content in the leaves that **were** exposed to 600 ppm continued to increase significantly throughout harvests 2 and 3.

Flavonoids are chemically important molecules in plants, as they contribute to the colour and scent of plants, and they also protect plants by acting as a signal molecule in times of stress. The molecule also exhibits strong therapeutic potential[2(1.21; thus, the increase in flavonoids in response to elevated CO_2 concentration may allude to the potential increase in therapeutic potential of the plant. This phytochemical group is also an important reactive oxygen species scavenger.[21]

Antioxidant activity

The leaves ol *P. afra* exhibited strong scavenging activity against DPPH (Table 4) and H_20_2 (Table 5). The efficacy of the extract was determined by the IC_{50}value. An extract was considered effective ii it was below the upper limit of 1O µg(ml.*n*

The leaf extracts that were exposed to 420 ppm exhibited moderate scavenging activity against DPPH, wtth all harvests exhibiting an IC'° valuebelow the upper limit, incontrast to thehexane anddichloromethane extracts that were consistently above the upper limit. No statistical

Table 1: The recorded weight and chlorophyllcontent of the leavesof *Portulacaria atra* after each harvest

	Harvest 1		Harvest 2		Harvest 3	
	420 ppm	600 ppm	420 ppm	600 ppm	420 ppm	600 ppm
Weight lll) of wel **leaves** jmean ± **SD)**	0.72 ± 0.43	1.08 ± 0.63	0.78 ± 0.03	1.65±0.71	0.89 ±0,07	1.79 ± 0.14
Chlorophyll content jmg/gFW)	5.95	5.99	5.90	6.83	6.44	6.68

Table 2: **A** heat map representing the presence of the different phytochemical groups in the leaves of *Portulacana afar* in the greenhouse (control) and those under 420 ppm and 600 ppm treatment

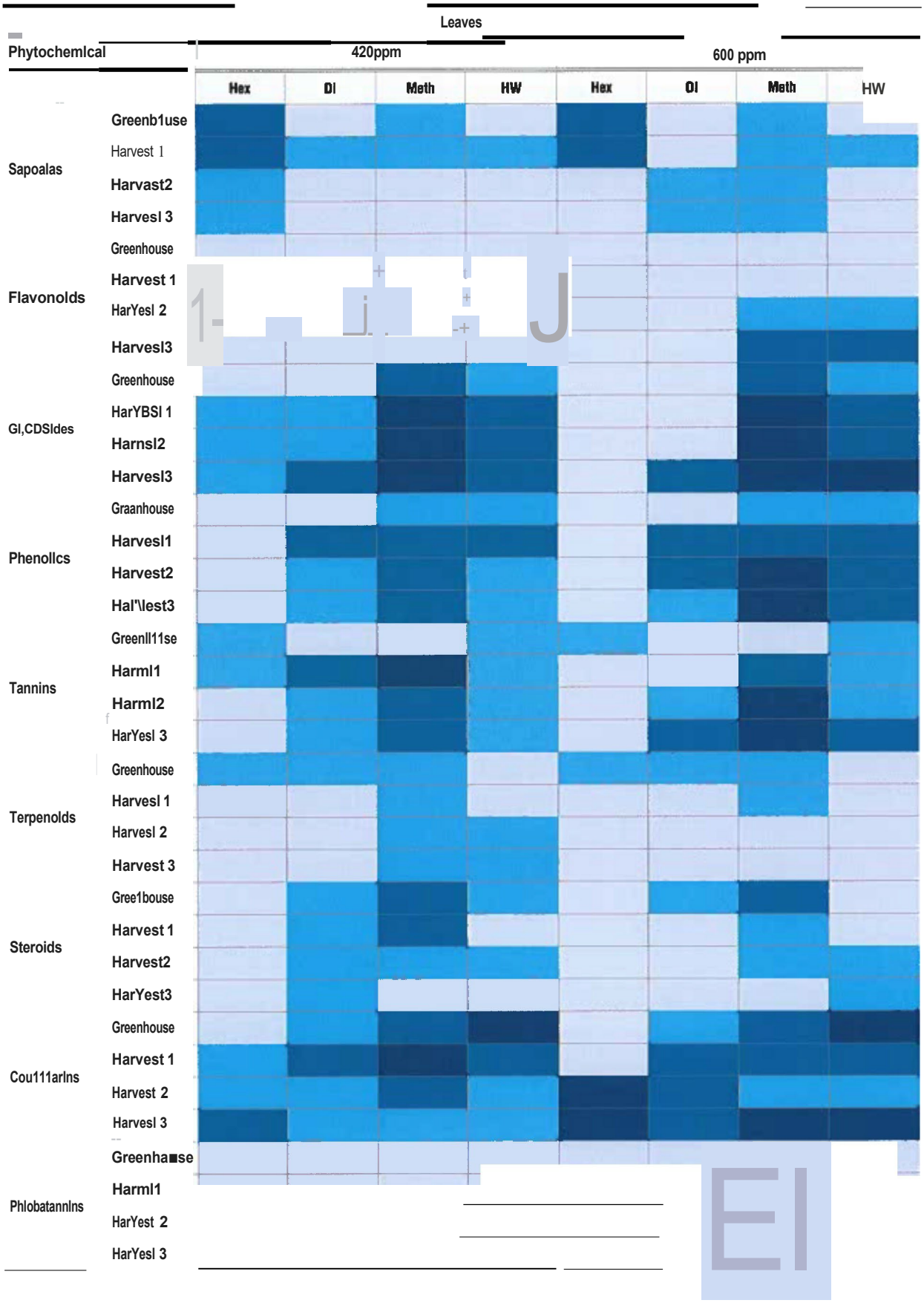


Table 2 continued...

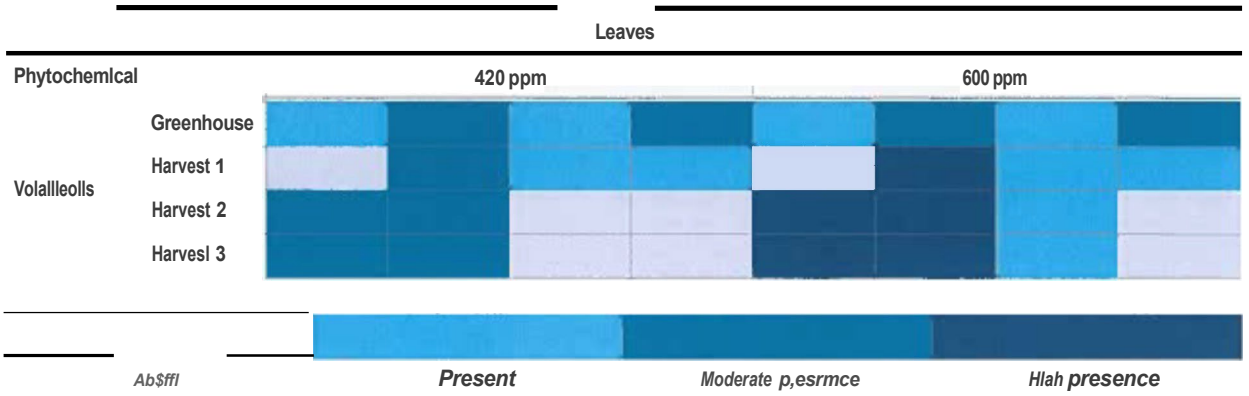


Table 3: Total llavonoid content in the leaf extracts of *Portulacariaafra* which were exposed lo 420 ppmand 600 ppm treatments in comparison to the control (greenhouse)

Leaves	Flavonold content (mg OE/g)							
	Methanol		Dlchloromethane		Hexane		Hot waler	
	420 ppm	600 ppm	420 ppm	600ppm	420ppm	600 ppm	420 ppm	600 ppm
Greenhouse	6.48 ± 0.32	6.48 ± 0.41	2.12 ± 0.87	2.12 ± 0.87	1.24 ± 0.54	1.24 ± 0.54	2.51 ± 0.43	2.43
Harvest 1	3.79 ± 0,19	9.38 ±0.45	1.62 ±0.39	6.33 ± 0.77	1.09 ± 0.37	2.54 ± 0.72	2.20 ±0.67	1265 :1:0.19*
Harvesl2	4.88 ± 0,43	13.80 ± 0.29•	3.45 ± 0.67	6.87 ± 0.68	1.92 ± 0.34	4.68 ± 0.69	3.79 :t 0.12	14 28 ± 0.31•
Harvesl3	4.79 ± 0.11	19.21 ± 0.1O•	3.69 ±0.34	4.65 ± 0.61	3.51 ± 0.39	7.20 ± 0.83·	4.31 ±0.19	1.81 ± 0.18*

•*sralistica/ly significantly different from lhe greenhouse extracts in a Tukey ad-hoc* 1es1

Table 4: The IC of the vanous leaf extracts exposed to 420 ppm and 600 ppm treatments 1n companson lo the control (greenhouse) against free radical DPPH

Leaves	IC value (µg/ml) ± SD							
	Meth nol		Dlchloromethane		Hexane		Hot water	
	420 ppm	600 ppm	420 ppm	600 ppm	420ppm	600 ppm	420 ppm	600 ppm
Greenhouse	4.7 ± 0.4	4.7 ±0.4	15.7±2.3	15.7 ± 2.3	14.8 :t 0.9	14.8±0.9	3.7 ± 0 3	3.7 ± 0.3
Harvest 1	1.7 ± 0.6	2.8 ± 0.9	18.3 ± 1.8	15.0:1:0.6	11.7±1.6	2.7 ± 1.3*	6.0 ± 0.1	4.3 ± 0.8
Harvest 2	3.3 ± 0.7	1.4 ± 0.2	11.4 ± 0.6	13.6 ± 1.5	10.4 ± 1.1	9.3 ±0.9*	9.4 ±0,3	3.1±02
Harvest 3	2.2 ± 0.6	2.1 ±0.4	129± 1.1	8.8 ± 2.5*	12.3 ± 1.2	12.0 ± 1.5	4.6 ±0,6	2.7 ± 0 5

•*sralis/Jcally significantly different from the greenhouse extracts in aTukey ad-hoc test* ($p < 0.05$)

Table 5: The IC,. of lhe various leaf extracts exposed to 420 ppm and 600 ppm treatments in companson to the control (greenhouse) against free radical H,O

Leaves	IC value (µg/ml) ± SD							
	Methanol		Dlch1oromethane		Hexane		Hot water	
	420 ppm	600 ppm	420 ppm	600 ppm	420 ppm	60 m	420 ppm	600 ppm
Greenhouse	0.3:t:0.2	0.3 ±0,2	18.7 ±0,8	18.7 ± 0.8	3.9 ± 0.8	3.9 ± 0.8	0.2 ±0,6	0.2 ±0,6
Harvest 1	4.1 ±0.4*	0.5 ± 0.6	24.6 ± 1.6	11.9 ±0.5	5.0 ± 0 9	5.3 ± 06	4.2 ± 0 2	0.8 ±0,2
Harvest 2	2 0 ± 0.3	0.9 ±0.3	13.9 ± 1.3	9.8 :t:0.8	5.8 ± 1 3	5.8 ± 0.9	4.4 ± 0.3	0 9 ±0,1
Harvest 3	0 4 ± 0.8	0.2 ±0.7	10.0 ± 1.7	7.8 ±1.3*	1.8 ± 0.9	7.2 ± 1.3*	0.7 ± 0 9	1 0 ±0 5

•*stalislically significantly different from lha greenhouse extracts in a Tukey acl-lloc test*

signtticance was determined foreitherextract exposed to420 ppmagainst the greenhouse extracts. The strongest scavenging activity was exhibited by the methanolic sample collected in harvest 1 (1.7±0.6 119/ml). The samples that were exposed to 600 ppm exhibited an overall stronger scavenging activity against DPPH than the 420 ppm-treated counterparts.

The methanolic and hot water extracts exposed to 600 ppm exhibited a consistenHy strong scavenging activity wilh IC_{50} values below 5 11o/ml across all three harvests. The strongest scavenging activity was exhibited by the methanolic sample collected in harvest 2. There was a general increase in scavenging activity across the various harvests in comparison to the greenhouse extracts. The hexanic leaf extracts that were exposed to 600 ppm exhibited significantiy stronger scavenging activity against the greenhouse extract (F4,21:4.3, p<0.05). The scavenging activity of the dichloromethane (harvest 3) and hexane (harvest 2) extracts was also significantly lower than that of the greenhouse leaf samples.

The leaves of *P.afra* exhibited agenerally stronger scavenging activity against H_20_2 than DPPH. The leaves exposed to 420 ppm exhibited a large variation in scavenging activity between the harvests. The dichloromethane extracts in harvest 1 had an ICso value of 24.6±1.6 119/ml, whereas that for harvest 3 was 10.0±1.7 1,19/ml. These large fluctuations can also be seen in the melhanolic and hot water 420 ppm extracts. There were also large differences between the 420 ppm extracts and their 600 ppm counterparts.

The leaves exposed to an elevated CO_2 concentration appear very effective as an antioxidant against $H_2 0_2$, with an IC that reached a low of 0.2±0.7 µg/ml in the methanolic extracts of harvest 3. There was a general increase in scavenging activity in leaves exposed to 600 ppm. The only exception to this finding was the hexane extracts that exhibited an IC_5 value of 3.9:t:0.8 µg/ml in the greenhouse extracts and reached 3.9±0.8 µg/ml in leaves exposed to 600 ppm after harvest 3.

The skin is particularly susceptible to reactive oxygen species attacks. This is aresult of the skin'sdirect exposure toUV radiation, environmental pollutants and high pressure of oxygen molecules.[25] Hence, the strong antioxidant activity exhibited by the plant may contribute to the relief of skin infections, which is what the plant is traditionally used for. The relatively steady antioxidant activity exhibited by the leaves indicates that the plant may retain its antioxidant activity in the future, despite the rising atmospheric COZconcentration. The phytochemicals that contribute to antioxidant activity in plants may also be effective at killing and hindering the growth of bacteria, viruses and fungi.[24]

Antimicrobial activity

The antimicrobial activity of the leaves of *P. atra* was tested against five microorganisms commonly associated with skin conditions. The antimicrobial activity of an extract was considered noteworthy if the MIC value was between 160 1,1g/ml and 1000 µg./ml.[2S]

The leaf extracts of *P. atra* mostly exhibited MIC values above 1000 µg/ml (Table 6), and hence, the antimicrobial activity is considered relatively low. Although the antimicrobial activity was low, ii was similar among the control and treated extracts. In samples exposed to an elevated CO_2 concentration, the extracts either showed a negligible increase in microbial activity (one serial dilution) or exhibited no change. A noteworthy result, however, was observed with *C. acnes.* The greenhouse plant extracts exhibited weak antimicrobial activity against *C. acnes,* but activity noticeably increased from an inhibitory concentration of 8000 µg/ml to 2000 1,1g/ml in plant extracts exposed to an elevated CO_2concentration.

C. acnes are pleomorphic rod-shaped bacteria and form a vital component of the skin's mlcroblota. This opportunistic pathogen has been idemified as a key component of acne vulgaris.[26] Thus, the leaves may continue to provide relief against this skin condition when the atmosphe<ic CO_2concentration reaches 600 ppm.

Conclusion

There is a concern amongst scientists that the rising atmospheric CO_2 concentration may affect the medicinal properties of plants. *P. afra* is a medicinally important plant used to treat skin infections. After exposure to an elevated CO_2 concentration, the plant exhibited a general increase in phytochemical presence. The antioxidant activity remained high in the methanolic and hot water extracts. The antimicrobial activity, while relatively weak, remained constant and the efficacy against *C. acnes* increased with increasing atmospheric CO_2 concentration.

The leaves of *P.afra* are resilient against exposure to an increased CO_2 concentration and may continue to provide relief against certain ailments in the future.

Thble 6: Mean inhibitory concentrations (MIC) of the leaf extracts of the control (greenhouse) and those which were subjected to ambient and elevated CO_2 concentrations (420 ppm and 600 ppm, respectively)

ant eldracl	Mean MIC values (pg/mLI					
	Gram-poslUve baclerfa			Gram-negallve bacteria		Yeasl
Sample	*S. IIIN8III*	*S. ,pldtrmfdls*	*C. acnu*	*P.aeruglnosa*	*IC.NtJJfltmU*	*C. alblcns*
Geenuse	4000	4000	**8000**	6000	2000	**8000**
420 (11	4000	4000	8000	4000	2000	8000
420(2)	4000	4000	4000	4000	2000	8000
420(3)	4000	4000	4000	4000	2000	8000
600 (1)	**4000**	4000	4000	**4000**	2000	8000
600 (2)	3000	4000	2000	**4000**	2000	8000
6 00(3)	2000	4000	2000	**4000**	2000	**4000**
Cipronoxacin (neiiative control)	0.625	0.625	1.25	1.25	0.625	N/A
Nystatin (p ositive controQ	NIA	NIA	N/A	NIA	NIA	2.50
Nejjative control	8000	>8000	8000	>8000	8000	>8000
CultUre control	**>8000**	>8000	**>8000**	>8000	>8000	**>8000**

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Competing interests

We have no competing interests to declare.

Authors' contributions

D.C.B.: Data collection; sample analysis; data curation; writing - the initial draft; writing - revisions. S.v.V.: Methodology; validation; writing - revisions; student supervision. I.M.R.: Conceptualisation; validation; writing - revisions; student supervision; funding acquisition.

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