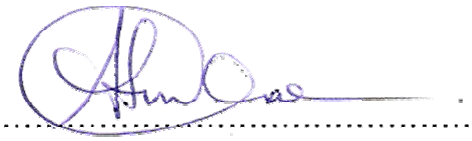
DECLARATION

I declare that this dissertation is my own, unaided work. It is being submitted for the Degree of Master of Science in the University of the Witwatersrand, Johannesburg. It has not been submitted before for any degree or examination in any University.



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26 JAN 2006

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ABSTRACT

An experimental study of the effects of a slag phase on mass transport in a one-fifth water model of a 100ton CLU-converter was conducted. The study was a follow up to earlier investigations conducted in the absence of a simulated slag phase. Kerosene (10% by volume) was used to represent the slag phase in the cold model experiments. The presence of a slag phase increased the mixing time of a tracer solution in the bath. The mixing time, defined at 99.66% bath homogeneity, was found to increase with bath height and a lowering gas flow rate. The functional relationship, $T_{mix} = 4.39Q^{-0.73}W^{0.24}H^{1.12}$, was established as expressing the effect of gas flow rate (Q), bath weight (W), and bath height (H) on the mixing time (T_{mix}). The mixing time increased by an average of 16.3% after slag inclusion. The mass transfer parameter $[(Re_{loc,r})^{0.25}(Re_t)^{0.32}]$ values obtained in the absence of a slag phase decreased by an average of 32.2% with slag inclusion. Calculated mass transfer coefficients increased with gas flow rate and a decrease in bath height. The relationship, $K \alpha Q^{0.08}$, showing derived mass transfer coefficient (K) dependence on the gas flow rate (Q) was established. The proportionality constant in the equation was observed to vary with bath height, gas flow rate and sample location. Contour maps representing variation of mass transfer coefficients in the bath regions were produced.

CONTENTS

LIST OF FIGURES

Figure **Page**

Figure **Page**

LIST OF TABLES

NOMENCLATURE

N_{Fr}	dimensionless, modified Froude number
ρ_g	gas density (kg/m^3)
ρ_l	liquid density (kg/m^3)
g	acceleration due to gravity (m/s^2)
d	diameter of the nozzle (m)
P	pressure of the gas (atm)
M	molecular weight of the gas (g mol^{-1})
I_t	time (s)
U_g	gas flow rate in real system (m^3/s)
U_{air}	gas flow rate in the full size water model (m^3/s)
$U_{g,nom}$	nominal velocity of the gas in the nozzle(m/s)
A	cross sectional area (m^2)
Q_m	gas flowrate operating in the water model of reduced size (m^3/s)
Q_{fs}	gas flowrate operating in the water model of full size (m^3/s)
λ	ratio of reduction.
μ	viscosity of a liquid (cP)
σ	surface tension between two liquids (dyne/cm)
T_{mix}	mixing time (s)
W	bath weight (tons)
v	velocity of the gas in the nozzle (m/ s)
KE	kinetic energy (J)
P_s	Stirring power (J/s)
$\dot{m}$	mass of the purging gas per unit time (kg/s)
ρ	density of the purging gas per unit time (g/m^3)
M	molecular weight of the purging gas (kg/mol)
R	ideal gas constant (J mol^{-1} K^{-1})
V	volume of the purging gas (m^3)
u	mean plume velocity (m/s)
L	liquid depth (m)
r	radius of vessel (m)

K	mass transfer rate constant (m/s)
b,m,y,z	experimentally determined coefficients
E_b	stirring energy density due to buoyancy (W/ton)
Q	gas flow rate (N m^3/s)
H	bath height/depth (m)
H_{wb}	water bath height (m)
E_k	stirring energy density (W/ton)
E_t	total mixing energy density (W/ton)
a	experimentally determined constant
n	gas stirring factor of gas flow rate
pH_{cs}	temperature compensated 'stable' pH value
pH_i	initial pH value of the solution before acid dosing
$pH_{99.66\%}$	pH value when the bath had reached 99.66% homogeneity level
D_o	mass diffusion coefficient
Sh	Sherwood number
Re	Reynolds number
$Re_{loc,r}$	local nominal Reynolds number
Re_t	local turbulent Reynolds number
Sc	Schmidt number
C^*_s	saturation concentration of acid (kg/m^3)
C_s	bulk concentration (kg/m^3)
μ	axial component of velocity (m/s)
v	radial component of velocity (m/s)
u_o	nominal gas velocity (m/s)
L_{mod}	the depth of liquid in model (m)
L_{fs}	depth of liquid in full-scale model (m)
T	temperature (K)