of the bulk composition rather than an effect of temperature, as commonly supposed.

The results of the present study indicate that the mechanism for the formation of pyrrhotite proposed by Davidson appears to be more feasible than that suggested by Ramdohr.

Young (1917) remarked on the fact that pyrrhotite was usually accompanied by an abnormal amount of chlorite. In order to see whether any such close relationship existed between the chlorite and pyrrhotite of the V.C.R., the chlorite and pyrrhotite contents (by weight) were plotted graphically (Figure Nos. 20 and 21). The scattered distribution of the points indicates that no obvious quantitative relationship exists between the chlorite and pyrrhotite in the V.C.R. at Venterspost and Libanon.
SPHALERITE.

At least three generations of sphalerite occur in the V.C.R. at Venterpost and Lebanon. The oldest generation occurs as tiny rounded grains of sphalerite completely encased inside grains of pyrite, which were obviously derived from an earlier horizon.

More generally, however, the sphalerite occurs as xenomorphic grains which replace earlier sulphides such as pyrrhotite and chalcopyrite. The first, and presumably the older, type displays reddish internal reflections under crossed nicols and occasionally contains oriented inclusions of chalcopyrite and also frequently shows (111) twinning of the larger grains. The second type exhibits honey-coloured internal reflections and does not contain any chalcopyrite. The difference in the colour of the internal reflections shown by the two types indicates that the darker variety is richer in FeS, being formed at a slightly higher temperature. Vokes (1957) has indicated that the iron content of honey-coloured sphalerites is of the order of 6%.

The grains of sphalerite from the V.C.R. were too small to permit the drilling-out of the grains from polished section for X-ray analysis without severe contamination. A sample of sphalerite was, however, obtained from the Bird Reef at Lipaardsvlei, and was examined by means of polished sections, and also chemically analysed for iron. The colour of the internal reflections shown by the Bird Reef specimens of sphalerite appeared to be virtually identical to that of the darker variety of
sphalerite present in the V.C.R. Fuller (1959) has shown, on the basis of unit cell measurements, that samples of sphalerite from the Bird Reef at Luipaardsvlei were most probably formed at about 450 degrees Centigrade.

Chemical analysis of the present sphalerite specimens from the Bird Reef showed an FeS content of 10.60% FeS. This figure, when interpreted in terms of the FeS-ZnS solvus as prepared by Kullerud (1953), suggests a temperature of formation of approximately 400 degrees Centigrade, no allowance being made for the effect of pressure and depth of burial. Barton and Kullerud (1959) further point out that equilibrium assemblages of sphalerite with pyrite will only give a minimum temperature of formation if the temperature is estimated from the solvus for the FeS-ZnS system.

These findings, supported by the fact that exsolved chalcopyrite has been found in sphalerite samples from the V.C.R. (Plate LIII), may be taken as an indication that the unmixing of the chalcopyrite from sphalerite occurred in the temperature range of 350-400 degrees Centigrade (Buerger 1934). This therefore represents a minimum temperature to which the V.C.R. must have been raised by the outpouring of the overlying Venterdorp lavas.
OTHER ORE MINERALS.

Chromite.

The chromite of the V.C.R. was found to occur in two different forms, either as dull black compact octahedra showing signs of rounding, or as brilliant black octahedra showing considerably less rounding and with a saccharoidal habit. These differences in the outward appearance of the two forms of chromite suggested a possible difference in composition. Accordingly, the chromite grains were handpicked and the two different forms were examined by the writer by means of X-ray diffraction using Copper K-alpha radiation.

Due to the high proportion of iron in the samples, the lines of the diffraction pattern showed considerable scattering. Despite this difficulty, the strongest lines, corresponding to d spacings at 2.51, 1.60, 2.08 and 2.95 Angstroms, were clearly visible. These spacings correspond very closely to those listed by Berry and Thompson (1962) for chromite with the composition \((\text{Fe}, \text{Mg})\text{Cr}_2\text{O}_4\). The two diffraction patterns obtained from the two varieties of chromite appeared to be identical. This indicates that, in spite of the differences in their outward appearance, the two forms of chromite occurring in the V.C.R. have the same composition.

Galena.

The galena of the V.C.R. was found to occur in three different generations. The oldest generation occurred as fine anhedral grains and stringers...
filling the interstices in fractured grains of uraninite, and must have originated as a result of the radioactive decay of the uraninite (Plate XV).

A second form of galena was found included in apparently ancient pyrite grains (Plate LIIF). The third and youngest generation of galena occurred either as isolated anhedral grains embedded in quartz or alternatively, was associated with sulphides such as pyrite, chalcopyrite and pyrrhotite. Where the galena was associated with the sulphides, it either took the form of irregular stringers filling any cracks in fractured sulphide grains, or actually replaced the sulphides.

Chalcopyrite and cubanite (I)

The chalcopyrite of the V.C.R. was usually found to occur as anhedral grains and masses replacing earlier pyrite and pyrrhotite (Plate LIV). A minute inclusion composed of chalcopyrite, pyrrhotite and gold, the three of which jointly formed a single grain with mutual grain boundaries suggesting simultaneous deposition, was found inside a large grain of apparently ancient pyrite. This pyrite showed no obvious signs of healed cracks. This evidence suggests that this minor phase of chalcopyrite must be older than the major phase of chalcopyrite.

A more detailed study of the chalcopyrite was made possible by the use of a stain first suggested by Wachromajew (1954) for the positive identification of chalcopyrite in sections suspected of containing
pentlandite, chalcopyrite and gold. The stain was prepared by mixing 100 volume strength hydrogen peroxide and concentrated ammonia (specific gravity 0.900) in the ratio of 1:1 by volume.

Unfortunately, no details for the use of the stain were given by Wachromajew. The present writer found that consistent results were only obtainable with this stain when a freshly prepared mixture was used.

It was also found that, prior to use, the mixture should be left to stand until it became warm, this stage being accompanied by effervescence. All vestiges of grease should be removed from polished sections before staining. The sections should not be immersed in the staining mixture for longer than 10 seconds at a time. The progress of the staining process should be constantly checked under a microscope.

The staining mixture works relatively slowly at first, and stains the chalcopyrite iridescent in shades of blue and red. At the same time, any features such as twinning or inclusions of exsolved cubanite are rendered clearly visible. Prolonged contact with the staining mixture results in the formation of a dull brownish film on the surface of the chalcopyrite, and renders the section totally unsuitable for micro-photography.

After etching by this method, the chalcopyrite in the present suite of samples was found to show twinning and also to contain numerous anisotropic, creamy-coloured laths of cubanite.
According to Schwartz (1927), the presence of such laths of exsolved cubanite is typical of high temperature deposits which had attained temperatures of at least 450°C. (Plates LV, LVI).

**Chalcopyrrhotite or Cubanite (II)**

According to Ramdohr (1960), chalcopyrrhotite has a brownish colour not unlike that of pyrrhotite, but with a somewhat brownish tinge, and is weakly anisotropic. Ramdohr further felt that this was the reason why chalcopyrrhotite had often been overlooked, or taken for pyrrhotite, or had been described on occasion as isotropic pyrrhotite or brownish isotropic chalcopyrite.

A mineral conforming to the characteristics listed by Ramdohr was found in the present suite of samples from the V.C.R. in intimate contact with undoubted pyrrhotite. After staining with a mixture of hydrogen peroxide and ammonia, the pyrrhotite, as may be seen in the attached photographs (Plates LVII, LVIII), was left completely unattacked, whilst the chalcopyrrhotite, which may be seen to occur as an aureole surrounding the pyrrhotite, was stained a dull brown. (Ramdohr was unable to suggest any specific etch reagent for the identification of chalcopyrrhotite, but felt that, in view of the overall tendency for chalcopyrrhotite to tarnish readily, it could be expected to be sensitive towards etch reagents.)

According to Edwards (1947), pyrrhotite dissolving in chalcopyrite at temperatures above 300°C
could be expected to react with the chalcopyrite and to give rise to an aureole of chalcopyrrhotite-chalcopyrite around the residual area of pyrrhotite. Borchert (1934) mentioned that the range of temperatures of solid solution of pyrrhotite and chalcopyrrhotite lay in the region between 350 and 550°C. Ramdohr (1960), however, merely mentioned that in those cases where the minerals were found associated with one another, the temperature of formation of the chalcopyrrhotite must have exceeded 240°C.

Unfortunately, no X-ray diffraction data are available for the positive identification of chalcopyrrhotite, which some authorities do not even accept as a distinct mineral. The optical properties of the mineral described above and the "text-book" aureole which it forms around the pyrrhotite, however, leave little doubt that this mineral is in fact chalcopyrrhotite (cubanite II).

Marcasite.

Only a single grain of marcasite was found during the present study. The grain occurred embedded in the centre of a large pyrite grain. At first inspection the overall colour of the marcasite grain closely resembled that of its pyrite host, but careful study revealed the presence of a slight brownish pleochroism. Under crossed nicols the grain showed intense anisotropism and could readily be distinguished from its pyrite host. Marked anisotropism is the
only reliable method known for distinguishing marcasite from pyrite, other than the use of X-ray diffraction. This latter technique could, however, not be used in the present instance, since the grain was too small to permit drilling out from its pyrite host without severe contamination.

**Cobaltite**

Cobaltite was found to be far more prevalent in the V.C.R. than was previously thought. It occurred as pinkish-white weakly anisotropic grains, which either had a rounded outline, or occurred as jagged irregular grains showing replacement by gold. (Plates XII, XI). A single pyritohedral crystal of cobaltite was also located. The association of cobaltite with gold, and its origin, has already been referred to under the heading of "Gold".

**Leucopyrite and loellingite.**

Leucopyrite occurred as anhedral intergrowths with pyrite (Plate LX). The mineral showed a distinct white colour and extreme hardness, and was strongly anisotropic, with greenish, purple and dark brown polarization colours. These polarization colours differed from those shown by a single grain of loellingite embedded in quartz, which was found to have yellow to deep blue polarization colours (Plate LX).

Both minerals showed the characteristic etch reactions listed for their identification by Short (1948).

The presence of leucopyrite in the V.C.R. at Venterpost has already been mentioned in an earlier
study by Davies (1949), so that little doubt concerning its identity exists.

Pentlandite.

Pentlandite was found to occur only as intergrowths in pyrrhotite in the V.C.R. Its nature and significance has already been discussed under the heading of "Pyrrhotite".

Other minerals present.

Various other minerals were identified in the present suite of samples from the V.C.R. by means of micro-chemical tests and/or optical studies. Unfortunately the identification of these minerals could not be verified by means of X-ray diffraction, as in most cases only minute isolated grains of these minerals were found. In addition, these grains were usually in intimate contact with, or embedded in, other minerals.

The presence of these minerals in the V.C.R. does not help in the elucidation of the mode of mineralization of the horizon. Accordingly, only the salient features of these minerals have been listed below.

Glaucodot.

A few strongly anisotropic small white grains with a faint pinkish tinge were found and identified as glaucodot. The glaucodot grains were embedded in pyrrhotite.
Osmiridium

A single rounded whitish grain showing a high degree of reflectivity and extreme hardness, and which was apparently negative to all etch reagents, was identified as osmiridium.

Skutterudite.

A few small tin-white coloured pyritohedra were found embedded in quartz, well below the plane of the polished section, and were tentatively identified as skutterudite.

Linnaeite.

A single grain conforming to the characteristics listed by Wachromejew (1954) for linnaeite was found. The mineral occurred as an overcapping on an ancient pyrite grain.

Arsenopyrite.

The presence of arsenopyrite in the pyritic and tuff beds of the V.C.R. has already been reported by Davies (1949). In the present suite of samples, only a single strongly-anisotropic diamond-shaped grain of arsenopyrite was found, embedded in a host of pyrite.

Millerite.

A cluster of accicular bronze-yellow crystals arranged in delicate radiating groups, not unlike millerite in appearance, was found embedded below the surface of a large quartz grain. These crystals were tentatively identified as millerite.
Niccolite.

The presence of niccolite in the horizon was established by micro-chemical analysis of a grain obtained from a heavy-mineral concentrate. No grains answering to the description of niccolite were detected in polished section.

Gersdorffite.

Polished section studies revealed the presence of a few white to steel-grey cubo-octahedra and striated pyritohedra embedded in quartz, which were tentatively identified as gersdorffite.

Covellite - Nacdicenite - Chalcoite - Bornite.

These four copper minerals were only found as very minute grains in close proximity to, or as inclusions in, chalcopyrite, and were identified solely on the basis of their optical characteristics.
PARAGENETIC SEQUENCE.

The paragenetic sequence of the minerals occurring in the V.C.R. was for the most part determined by the classical criteria suggested by workers such as Bastin and others (1931) and Schwartz (1942).

In those cases where the order of deposition could not be determined with certainty from polished section studies alone, criteria such as exsolution features and geochemical evidence were used as geological thermometers, in the manner suggested by Ingerson (1955) and McKinstry and Kennedy (1957).

The sequence of deposition of the minerals with respect to time shown in the accompanying table indicates only the general trend, and does not reflect minor and obviously localized departures from the overall pattern of mineralization.
**Paragenetic Sequence of Minerals in the V.C.R. at Velterspost and Lipanon.**

<table>
<thead>
<tr>
<th>Mineral</th>
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<tr>
<td>Gold</td>
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<tr>
<td>Uraninite</td>
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<td>Osmiridium</td>
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<td>Sphalerite</td>
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<td>Galena</td>
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<td>Carbon</td>
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<td>Glauconite</td>
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<td>Marcasite</td>
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<td>Argenopyrite</td>
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<tr>
<td>Leucopyrite</td>
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<td>Lochlingite</td>
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<td>Gersdorffite</td>
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<td>Skutterudite</td>
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<td>Niscolite</td>
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<td>Anatase</td>
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<td>Covellite</td>
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<tr>
<td>Neo-digennite</td>
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**Vertical axis:** Order of deposition.

**Horizontal axis:** Time.
ECONOMIC IMPLICATIONS:

The results of the present study suggest that the following factors may be responsible for "tailing losses" (incomplete recovery of the gold) experienced during the extraction of the gold from the V.C.R. at Venterspost and Libanon:

1. Coatings on gold particles a) iron oxides
   b) silver

2. Fine size range of the gold

3. Presence of phyllosilicates in the ore

1. Coatings:
   a) Iron oxides:

   The V.C.R. ore at Venterspost and Libanon contains appreciable amounts of pyrrhotite and an unstable variety of pyrite. Both these minerals oxidize readily, with the concomitant liberation of ferrous iron ions. This ferrous iron then oxidizes further and coats the particles of gold in the ore with ferric oxide, thereby rendering the gold non-amenable to cyanidation. In order to minimize tailing losses, therefore, the V.C.R. ore should not be stockpiled for any length of time.

   b) Silver:

   The presence of coatings of silver on a few gold grains found during the present study, although probably of academic interest, may possibly account for the presence of isolated bright particles of gold in the "tails". It is a well-known fact that the rate of dissolution of silver in a cyanide solution is considerably slower than that of gold, and accordingly, where
gold grains are coated with silver, the time normally allowed for the cyanidation of ores would probably only be sufficient to allow for the removal of the silver coating and partial dissolution only of the underlying gold grain.

2. Fine size range of the gold.

As can be seen from Figure No. 1, most of the gold in the V.C.R. was found to be of a very fine nature, with over 80% of the gold grains being smaller than 90 microns in size, whilst 37% of the gold grains were less than 15 microns in size. The grains in the 1-15 micron range were almost all chatted with quartz and phyllosilicates. Such grains would probably not be liberated completely during the crushing process used on the mines, and would therefore possibly not be amenable to cyanidation.

3. Presence of phyllosilicates in the ore.

Further tailing losses may possibly be caused by the presence in the V.C.R. ore of large amounts of phyllosilicates, which have a tendency to clog the pores of the filters used to remove the gold-cyanide complex from the pulp.

Variations in the apparent fineness of gold with the tenor of the ore.

A further factor which should be borne in mind when considering the discrepancy which often exists between "call" and recovery figures of gold, is the variation in the apparent fineness of the gold with the
tenor of the ore. These variations are shown graphically in Figures nos. 7 and 8. These graphs show a progressive increase in the apparent fineness of the gold with increasing gold value of the ore until a critical point is reached, after which the apparent fineness of the gold remains reasonably constant. It is therefore felt that, when estimating "call" figures, the use of one fixed overall value for the silver content of the gold may introduce errors. This is especially the case when some of the ore is of such a grade that it lies below the critical point on the graph, in the region where the apparent fineness of the gold varies rapidly with the tenor of the ore. This question clearly needs further investigation on a scale beyond the scope of the present survey.
CONCLUSIONS.

The origin of the gold in the Witwatersrand Banket has been a subject of discussion and heated controversy ever since gold was first discovered on the Witwatersrand in the latter part of the last century. There are two major schools of thought concerning the origin of the gold: one school envisages a hydrothermal origin for the gold, whilst the other supports a placer origin. The theory apparently receiving the most support at present is a variation of the placer theory known as the "modified placer theory". This hypothesis envisages the recrystallization and/or solution followed by redeposition of previously alluvial gold, with the assumption that these processes took place more or less in situ.

With regard to the origin of the gold in the V.C.R., the results of the present study at Venterapost and Lebanon seem, on first inspection, to be somewhat inconclusive. The majority of the gold grains found in the V.C.R. were hackly and irregular in outline, a mode of occurrence which seems to favour a purely hydrothermal origin for the gold, but on the other hand, strong support for an original placer origin of the gold is provided by the small cylindrical grains of gold which were isolated from apparently ancient pyrite. These cylinders appear to be identical to those illustrated by Dunn (1929) for gold of undoubted detrital origin from Australia. They probably owe their preservation in the V.C.R. to their pyrite host, which
presumably prevented the grains from undergoing solution and subsequent redeposition, which appears to have been the fate of the major portion of the gold.

Both of these forms of gold existing in the V.C.R. are, in fact, those which are to be expected in terms of the modified placer theory.

If the modified placer theory is valid, then the source of the gold in the V.C.R. must be some horizon at present underlying the V.C.R. In this context, it is interesting to note that studies carried out on the zircons of the Main Reef and the overlying V.C.R. showed such close similarities as to suggest a common source for the zircons present in both horizons.

Davies (1949), on the other hand, has presented evidence which, in his view, would rule out the Main Reef as a source of the gold in the V.C.R. His argument was based on the following grounds: firstly, that the grains of gold in the V.C.R. were coarser than those in the underlying Main Reef, which was contrary to what would have been expected if attrition and transportation losses had occurred, and secondly, that the fineness values of the gold in the V.C.R. were practically identical to those of the gold in the Main Reef. This similarity in the finenesses is not in accordance with the trend envisaged by Lindgren (1913), who found that the "fineness" of alluvial gold apparently increased downstream from the lode source, and attributed this feature to the preferential leaching of silver relative
to gold by water in a process of erosion and stream transport.

The findings of the present writer are in agreement with those of Davies both with regard to the coarseness of the grains of gold in the V.C.R. relative to those in the Main Reef and with regard to the similarity of the upper apparent fineness values of the two horizons. The writer feels, however, that these facts may have an alternative interpretation, which would not necessarily result in the Main Reef being ruled out as a possible source of the gold in the V.C.R.

Firstly, Lindgren's explanation of the reason for the fineness gradient found by him in alluvial gold may not be valid in the present context: experiments carried out by the present writer have clearly shown that even nitric acid is unable to remove silver from alloys of gold and silver whose fineness exceeds 615 parts per thousand. (It should be noted that the major portion of the gold examined from both the V.C.R. and the Main Reef had apparent fineness values well above 615.) Accordingly, Davies' (1949) objection concerning the similarity in the apparent fineness values of the two horizons appears to be irrelevant. Further, it is felt that the fact that the grains of gold in the V.C.R. are coarser than those in the Main Reef may possibly be accounted for if it is assumed that solution of the smaller gold grains derived from the Main Reef took place, followed by their redeposition in the V.C.R. as coarser hackly particles and crystals. Evidence to this effect is afforded by the presence of a few minute
euheiral and subhedral crystals of gold with bright facets showing no signs of attrition, observed in polished section. More detailed polished section studies further revealed the presence of various micro-structures in the gold grains themselves after the grains had been etched. These microstructures and the irregular outline shown by the gold grains (to which reference has already been made) strongly suggest that the gold was apparently recrystallized more or less in situ, and that it has not undergone any significant degree of transportation since its deposition in the V.C.R.

The present writer feels that the basic requirements of the modified placer theory are met by the presence in the V.C.R. of both the cylinders of apparently original detrital gold, which has obviously undergone recrystallization and redeposition in situ, as shown by the micro-structures present in the gold. It is also felt that the Main Reef may be considered a probable source of the gold in the V.C.R.

The temperature of mineralization of the V.C.R. is attested to by various geological thermometers, which are briefly listed as follows:

1. Sphalerite host showing blabs of exsolved chalcopyrite. Temperature range 350-400°C (Buerger, 1934).


These geological thermometers, it is felt, indicate that the temperature of the V.C.R. was not raised to very much more than 450°C as an upper limit as a result of the outpouring of the overlying Ventersdorp Lavae.
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# ABBREVIATIONS USED IN MICROGRAPHS

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<td>Sphalerite</td>
<td>ZN</td>
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<tr>
<td>Uraninite</td>
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Plate I

Steeply dipping V.C.R. with central band of glassy quartzite. Note the wide variation in the size of the pebbles and cobbles.

Plate II

V.C.R. ore containing an angular fragment of banded chert. Note the dark chlorite matrix and the replacement by chlorite on the periphery of the quartz pebbles.
Hackly grains of gold and rounded zircon grains.

Cylindrical and hackly grains of gold.
Etched gold grains apparently showing twinning.

Etched grain of gold. Note the sharp boundaries of the segments.
Etched grain of gold showing minute crystals of included gold and "islands" of pyrrhotite.

Adjacent portions of an etched grain of gold showing differences in orientation, with suggestion of cubic structure.
Gold replacing pyrrhotite.

Anhedral stringer-like grains of gold in a quartz and chlorite matrix.
Gold replacing irregular grain of cobaltite.

Abraded grains of cobaltite and pyrite.
Gold clearly filling cracks in uraninite and "carbon" host. Note also the presence of galena grains in the cracks in the uraninite.

Pyrite grain showing rounded and cylindrical inclusions of gold. Dark spots are inclusions of quartz.
Gold replacing uraninite. Note the presence of fine stringers of "carbon" cementing the uraninite grain, and fine specks of included galena.

Poorly polished grain of native silver (?) embedded in eroded phyllosilicate matrix.
Irregular grain of apparently once plastic "carbon" wedged between rounded pyrite grains.

Irregular grains of "carbon" moulded about, and containing inclusion of, pyrite, and also showing replacement by pyrrhotite.
Irregular "carbon" grain showing fibrous habit. Note the haphazard orientation of the fibres.

"Carbon" moulded around pyrite. Note the darkening of the "carbon" where in contact with the pyrite.
"Carbon" with inclusions of shattered subhedral uraninite.

Fragments of euhedral uraninite in "carbon" surrounded by shattered pyrite.
Uraninite with thin shell of "carbon" transected by veinlets of "carbon". Note the pyrrhotite inclusion and grain of gold on the periphery.

Uraninite (black) stained with HF apparently showing (111) cleavages, with fractures filled by "carbon". Note the degree of replacement by gold on the periphery.
Subhedral uraninite (black) after staining with HF, cemented by fine grey veinlets of "carbon".

Zircons showing varying degrees of rounding.
Plate XXVII

Geniculate zircon grain.

Plate XXVIII

Zircon grain showing outgrowth.
Zircon grain showing overgrowth.

Zonal zircon with a euhedral core.
Plate XXXI

Zircon with randomly oriented rounded and rod-like inclusions.

Plate XXXII

Geniculate zircon inclusion in gold.
Grain of leucoxene and pyrite grains.

Cluster of subhedral rutile.
Scaly mixture of chlorite and sericite.

Scaly mixture of chlorite and sericite replacing quartz.
Fine flakes of sericite replacing quartz.

Quartz. Note the fine black inclusions of "dust".
Quartz grains showing strain.

Radial pyrite.
Rounded pebble, probably originally a ferruginous sandstone rich in quartz, which has been changed into pyrite.

Rounded pebble of pyritized sediment.
Pyrite grains formed from smaller cemented angular grains, with inclusion of pyrrhotite.

Authigenic zonal pyrite.
Banded porous pyrite -- probably a pyritized sediment.

Younger zonal pyrite which has grown around older "clear" pyrite.
Pyrite crystal with an overcapping of younger zonal pyrite.

Juvenile stringer-like pyrite.
Fragments of pyritized ferruginous sandstone? 
Dark inclusions are quartz.

Subhedral pyrrhotite replacing radial pyrite.
Pyrrhotite showing beta lamellae.

Sphalerite showing blebs of exsolved chalcopyrite. The white mineral is pyrite.
Galena inclusions in shattered pyrite.

Chalcopyrite replacing pyrrhotite (chalcopyrite repolished after etching).
Stained chalcopyrite showing twinning, exsolved laths of cubanite and inclusions of pyrrhotite.

Stained chalcopyrite showing laths of exsolved cubanite. Other mineral is pyrite.
Aureole of chalcopyrrhotite enclosing a pyrrhotite core. Other minerals are stained chalcopyrite and pyrite.

Pyrrhotite with aureole of chalcopyrrhotite.
Plate LIX

Leucopyrite intergrown with, and rimming, pyrite.

Plate LX

Lecelllingite.
TABLE I.
SAMPLING SITES, ELEVATIONS AND ASSAY RESULTS.

<table>
<thead>
<tr>
<th>SAMPLE</th>
<th>VENTERSPROG</th>
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<th>SILVER DWT/TON</th>
<th>APPARENT FINENESS</th>
<th>ELEVATION FEET</th>
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TABLE II.

SAMPLING SITES, ELEVATIONS AND ASSAY RESULTS.

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**TABLE III.**

**SAMPLING SITES, ELEVATIONS AND ASSAY RESULTS.**

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<td>14-16 Dr. E 10423 + 115½' H.G.</td>
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<td>16-17 Dr. E 10425 + 93½' L.G.</td>
<td>0.67</td>
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<td>VENTERSPOST SILVER DWT/TON</td>
<td>MAIN REEF APPARENT FINENESS</td>
<td>ELEVATION FEET</td>
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<td>---------------------------</td>
<td>----------------------------</td>
<td>------------------------------</td>
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**Table V.**

**Sampling Sites, Elevations and Assay Results.**

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<td>SILVER DMT/TON</td>
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-187-

**TABLE VII.**

**TABLE SHOWING FINENESS VALUES OF SAMPLES USED IN**

"FREE SILVER" DETERMINATION.

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<th>AVERAGE APPARENT FINENESS</th>
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<tr>
<td>Moisture %</td>
<td>3.84</td>
<td>4.00</td>
<td>1.20</td>
<td>0.80</td>
<td>3.70</td>
<td></td>
</tr>
<tr>
<td>Volatile Matter %</td>
<td>16.82</td>
<td>18.76</td>
<td>21.22</td>
<td>15.54</td>
<td>22.65</td>
<td></td>
</tr>
<tr>
<td>Fixed Carbon %</td>
<td>61.84</td>
<td>65.34</td>
<td>65.00</td>
<td>69.16</td>
<td>53.25</td>
<td></td>
</tr>
<tr>
<td>Fixed Carbon (A.S.T.M.)%</td>
<td>80.90</td>
<td>79.30</td>
<td>77.80</td>
<td>83.80</td>
<td>71.80</td>
<td></td>
</tr>
<tr>
<td>Ash %</td>
<td>17.50</td>
<td>11.90</td>
<td>12.58</td>
<td>14.50</td>
<td>21.30</td>
<td></td>
</tr>
<tr>
<td>Sulphur %</td>
<td>2.32</td>
<td>2.50</td>
<td>4.25</td>
<td>2.53</td>
<td>1.04</td>
<td></td>
</tr>
<tr>
<td>Calorific Value</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>E.Th.U.</td>
<td>11710</td>
<td>12700</td>
<td>13400</td>
<td>12910</td>
<td>11280</td>
<td></td>
</tr>
<tr>
<td>Calorific Value</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>(A.S.T.M.) E.Th.U.</td>
<td>14530</td>
<td>14660</td>
<td>15730</td>
<td>14550</td>
<td>14720</td>
<td></td>
</tr>
</tbody>
</table>

1. Luipaardsvlei Est. & G.M. Co. Ltd. - Bird Reef Series
2. Vogelstruisbult G.M. Co. Ltd. - Main Reef
3. Vogelstruisbult G.M. Co. Ltd. - Kimberley Reef Series
4. Doornfontein G.M. Co. Ltd. - Carbon Leader
5. West Driefontein G.M. Co. Ltd. - Carbon Leader

# Table IX

Table showing Zircon Elongation/Frequency Relationships.

<table>
<thead>
<tr>
<th>Elongation</th>
<th>V.C.R. Frequency per cent</th>
<th>Main Reef Frequency per cent</th>
</tr>
</thead>
<tbody>
<tr>
<td>1.00 - 1.25</td>
<td>0.00</td>
<td>1.06</td>
</tr>
<tr>
<td>1.25 - 1.50</td>
<td>8.60</td>
<td>4.62</td>
</tr>
<tr>
<td>1.50 - 1.75</td>
<td>18.28</td>
<td>15.25</td>
</tr>
<tr>
<td>1.75 - 2.00</td>
<td>18.28</td>
<td>13.82</td>
</tr>
<tr>
<td>2.00 - 2.25</td>
<td>30.10</td>
<td>30.50</td>
</tr>
<tr>
<td>2.25 - 2.50</td>
<td>13.97</td>
<td>17.72</td>
</tr>
<tr>
<td>2.50 - 2.75</td>
<td>5.38</td>
<td>9.22</td>
</tr>
<tr>
<td>2.75 - 3.00</td>
<td>1.08</td>
<td>3.89</td>
</tr>
<tr>
<td>3.00 - 3.25</td>
<td>2.15</td>
<td>2.12</td>
</tr>
<tr>
<td>3.25 - 3.50</td>
<td>1.08</td>
<td>0.36</td>
</tr>
<tr>
<td>3.50 - 3.75</td>
<td>1.08</td>
<td>0.00</td>
</tr>
<tr>
<td>3.75 - 4.00</td>
<td>0.00</td>
<td>0.36</td>
</tr>
<tr>
<td>4.00 - 4.25</td>
<td>0.00</td>
<td>0.36</td>
</tr>
<tr>
<td>4.25 - 4.50</td>
<td>0.00</td>
<td>0.36</td>
</tr>
<tr>
<td>&gt; 4.50</td>
<td>100.00</td>
<td>100.00</td>
</tr>
</tbody>
</table>
FIGURE 1.

SIZE DISTRIBUTION OF GOLD PARTICLES RECOVERED FROM 18 SAMPLES OF PULVERISED ORE.

FIGURE 2.

SIZE DISTRIBUTION OF GOLD PARTICLES RECOVERED FROM AN AREA OF COARSE BUCKSHOT PYRITE.
FIGURE 3.

Graph showing gold/pyrrhotite relationship

VENTERSDORP CONTACT REEF
VENTERSPOST

GOLD DWT/TON

PYRRHOTITE PERCENT
FIGURE 4.

GRAPH SHOWING GOLD/PYRRHOTITE RELATIONSHIP

VENTERSDorp CONTACT REEF
LIBANON.

PYRRHOTITE PERCENT
FIGURE 5.

GRAPH SHOWING GOLD/URANINITE RELATIONSHIP

VENTERSDORP CONTACT REEF

VENTERSPOST

COLD DWT/TON

\( \epsilon U_3O_8 \) LBS/TON
FIGURE 6

Graph showing gold/uraninite relationship.

VENTERSDORP CONTACT REEF
LIBANON.
FIGURE 7

Graph showing relationship between apparent fineness of gold and tenor of ore.

VENTERSDORP CONTACT REEF

VENTERSPOST

FIGURE 8

Graph showing relationship between apparent fineness of gold and tenor of ore.

VENTERSDORP CONTACT REEF

LIBANON
**Figure 9.**

Graph showing relationship between apparent fineness of gold and tenor of ore.

*Venterstorp Contact Reef*

*West Driefontein*

**Figure 10.**

Graph showing relationship between apparent fineness of gold and tenor of ore.

*Main Reef, Venterstpoet*
FIGURE 11

GRAPH SHOWING RELATIONSHIP BETWEEN APPARENT FINENESS OF GOLD AND TENOR OF ORE

MAIN REEF
LIBANON

FIGURE 12.

GRAPH SHOWING RELATIONSHIP BETWEEN FINENESS OF GOLD-SILVER ALLOY AND ABILITY TO RESIST REMOVAL OF SILVER BY NITRIC ACID

PERCENTAGE OF TOTAL AMOUNT OF SILVER REMOVED
CLASSIFICATION OF CARBONACEOUS MATERIALS

SOLUBILITY IN CARBON DISULPHIDE

BITUMENS

- SOLUBLE
  - LIQUID
    - PETROLEUM
  - SOLID
    - FUSIBLE
      - ASPHALTITES
        - OZOCERITE
        - MONTAN WAX
        - HATI-BETITE
      - ASPHALT
        - BERMUEDEZ PITCH
        - TABBYTE
        - ARGULITE

- INSOLUBLE
  - PYROBITUMENS
    - OXYGEN FREE
      - ASPHALTIC
        - ELATETERITE
        - WURZILITE
        - ALBERTITE
        - IMPSONITE
    - OXYGEN CONTAINING
      - NON ASPHALTIC
        - PEAT
        - LIGNITE
        - BITUMINOUS
        - ANTHRACITE

(AFTER HUNT STEWART AND DICKEY 1954)
HISTOGRAM SHOWING SIZE DISTRIBUTION OF URANINITE PARTICLES RECOVERED FROM PULPED ORE.
FIGURE 15

GRAPH SHOWING RELATIONSHIP BETWEEN ZIRCON AND URANINITE

VENTERSDORP CONTACT REEF
VENTERSPOST

NUMBER OF ZIRCONS COUNTED

$eU_3O_8$ LBS/TON
FIGURE 16

GRAPH SHOWING RELATIONSHIP BETWEEN ZIRCON AND URANINITE

VENTERSDORP CONTACT REEF

LIBANON.
**FIGURE 17**

ZIRCON TYPES - (AFTER DANA 1909)

1. 2. 3. 4. 5. 6.

**FIGURE 18**

ZIRCON FREQUENCY\ELONGATION DIAGRAM

**LEGEND**

BLUE: MAIN REEF SUB NIGEL
RED: V.C.R. VENTERSPOST
FIGURE 19

ZIRCON FREQUENCY/ELONGATION DIAGRAM FOR COARSE GRANITE A 124 (AFTER WYATT 1954)
FIGURE 20.

GRAPH SHOWING PYRRHOTITE CHLORITE RELATIONSHIP

VENTERSDORP CONTACT REEF
VENTERSPOST

FIGURE 21

GRAPH SHOWING PYRRHOTITE CHLORITE RELATIONSHIP

VENTERSDORP CONTACT REEF
LIBANON
LIBANON GOLD MINING COMPANY LIMITED

PLAN OF THE UNDERGROUND WORKINGS AS AT 30TH JUNE, 1962

AREA = 36443479 CLAIMS

REFERENCE
Development & Stoping on Ventersdorp Contact Reef
Development & Stoping on Main Reef
Development on Carbon Leader
Development off Reef
Ore Reserve
Sub-outcrops of Reefs
Dikes

LIBANON 283-IQ

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Development & Stoping on Main Reef
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Development on Carbon Leader
Development off Reef
Ore Reserve
Sub-outcrops of Reefs
Dikes
TRANSVERSE SECTION THROUGH Nº1 SHAFT LOOKING NORTH

REFERENCE:
Ventersdorp Contact Reef  Main Reef  Carbon Leader
Faults  Dykes

KEY PLAN
showing
LIBANON G. M. CO. LTD.
in relation to neighbouring mines.
Scale 1 inch = 1 mile.

For Key Plan see overleaf.
Author  Rahden H V R
Name of thesis  A detailed mineralogical investigation of the Ventersdorp contact Reef at Venterspost and Libanon Gold Mines, West Rand, with special reference to the mode of mineralization of this horizon  1963

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