

CHAPTER SEVEN

MINERALIZATION

OXIDIZED ZONE:

The zone of oxidization extends down to a depth of approximately 120 feet below the surface over all the area mined. The payable rubble and oxidized ore have long since been mined out. Thus very limited material is available for examination. The oxidized ore was a very weathered and leached banded jaspilite, containing a high proportion of iron minerals, and was comparatively rich in gold in some places. According to records, the average grade of the ore mined before 1900 was above 12 dwts/ton. Where gossan was prominent favourable gold values were found. Where the weathered banded ironstone passed into a more massive cherty jaspilite along its strike, the values decreased.

No detailed examination of the oxide minerals has been made. They appear to be quartz and hydrated iron oxides. Thus weathered outcrops show only minor magnetic effect. One uncommon mineral occurs, however, which has been identified as vivianite-vauxite. Its occurrence and features are as follows.

In many places along the jaspilite outcrops, but best developed in the vicinity of the Magano section, a greenish yellow mineral is present. It occurs as small discontinuous flat lenses up to 10 millimetres thick with a horizontal attitude independent of the local rock. These lenses occur for the first five

or six . . / -

ARSENOPYRITE-ROLLINGITE:

Since the name "Arsenopyrite" is used in all the literature on Vubachikwe Mine - as well as on the property itself - it has been retained throughout the text, except partially in this particular section.

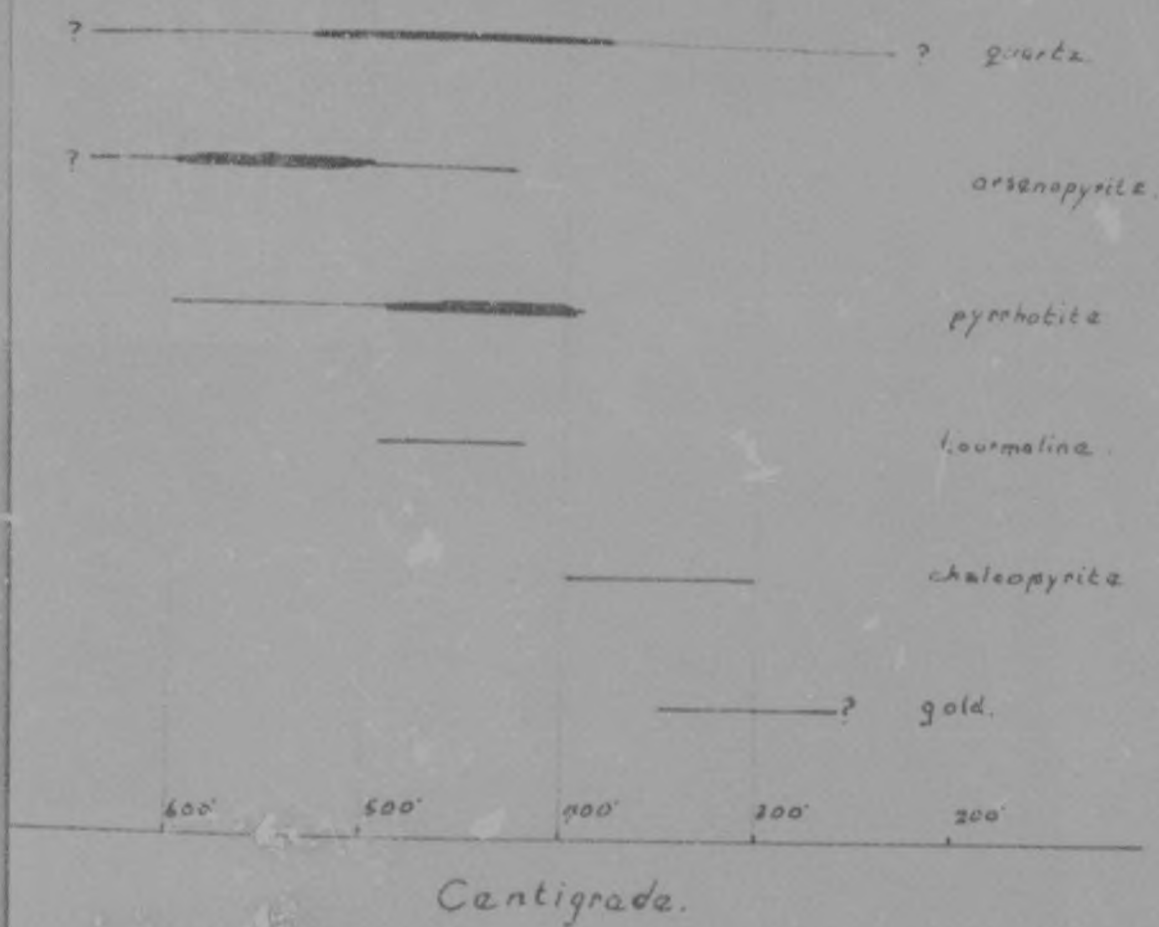
Arsenopyrite is the commonest sulphide of the ore, amounting to about 7% of the ore by volume. This estimate was reached by careful visual examination of the ore and of concentrate samples. The mineral occurs disseminated in the jaspilite. It ranges from small anhedral aggregates to euhedral crystals up to 15 millimetres long, and about 1.5 millimetres thick. Freshly broken crystals have a highly metallic lustre which tarnishes rapidly to a dull grey-white.

Under the microscope, the arsenopyrite is opaque in transmitted light, and bright white in reflected light. It is a very weakly pleochroic, the only change noticed being a slight variation in the intensity of colour on rotating the stage. Under crossed nicols it is weakly anisotropic, changing from a steel blue-grey to a brownish-grey.

Examination of the arsenopyrite in polished sections shows that it occurs with a fairly wide range of textures. The impression gained is that it was the earliest of the sulphides to crystallise, and that this crystallisation lasted through a considerable time, and probably over quite a wide temperature range. On the edges of ore-bodies - i.e., at the limits of impregnation of the

solutions . . . / -

Paragenetic Chart.



or six feet below the surface and do not continue below this depth. Due to the greenish colour contrasting with the reddish brown of the limonitic soil and banded ironstone, the mineral was initially examined for the presence of radio active material. The occurrence has been noted by the Geological Survey of Southern Rhodesia and was described by Phaup*. The mineral was identified by the Geological Survey of Southern Rhodesia as vivianite, an obscure hydrated iron phosphate with the formula $\text{Fe}_3 \text{P}_2 \text{O}_8 \cdot 8\text{H}_2\text{O}$. In the present investigation, however, it has been found that the mineral is almost amorphous. It has a massive clayey texture, unlike the fibrous flexible texture of vivianite. Analysis by Xray diffraction shows the mineral to belong to the Vauxite group, another series of hydrous iron phosphates. The variety examined corresponds most closely to the semi-amorphous mineral delavauxite with the formula $2 \text{Fe}_2 \text{O}_3 \cdot \text{P}_2 \text{O}_5 \cdot 9\text{H}_2 \text{O}$. It is a secondary mineral resulting from soil water action, the formation being similar to that of calcrete. The origin of the phosphate content is obscure, since no phosphate minerals are known to be present in the local rocks. It is possible, however, that phosphate ions may be associated with carbonates which are so prevalent in the rocks of this area.

SULPHIDE ZONE:

In the sulphide zone, extending below the oxidized zone, the ore is found to contain the following minerals, each of which is described over page:-

. . . / -

*"Notes on the Marand Mine" (22-11-52)

solutions - the magnetite was only partially replaced. Within the orebodies, unreplaced magnetite does occur - but uncommonly. Thus it may be concluded that the first sulphide crystals formed in an iron-rich environment and actually absorbed the iron by replacing magnetite with arsenopyrite.

Arsenopyrite occurs in its commonest form in well formed crystals. The most common form observed is of the pseudo orthorhombic prism 110. Also common are vertically flattened crystals as well as columnar, straight, divergent, granular and compact. Twinned groups and combined crystal forms are not uncommon.

In the early stages of crystallisation of the arsenopyrite, the earliest perfect euhedra was formed. Within the comparatively wide range of temperature of formation of arsenopyrite, tourmaline also crystallised in close association with the arsenopyrite. The larger arsenopyrite crystals seem to have suffered displacement and fracturing and this indicates that at least some of the sulphide is older than the tourmaline, which probably crystallised simultaneously with the last of the arsenopyrite. See Plates Nos. 24, 25, 26, 27 and 29.

Numerous aggregates of small arsenopyrite crystals occur. They all exhibit a mutual growth interference and suggest an increase in the rate of deposition as the temperature fell. Pyrrhotite is seen in mutual relationship with these aggregates and at approximately this stage began to crystallise together with the arsenopyrite. See Plates Nos. 28, 30 and 33.

The arsenic bearing . . / -



PLATE NO. 22 : Garnet and Carbon in Slate.
Opaque black is mostly quartz,
coated with opaque graphite.
Plane polarised x 50.

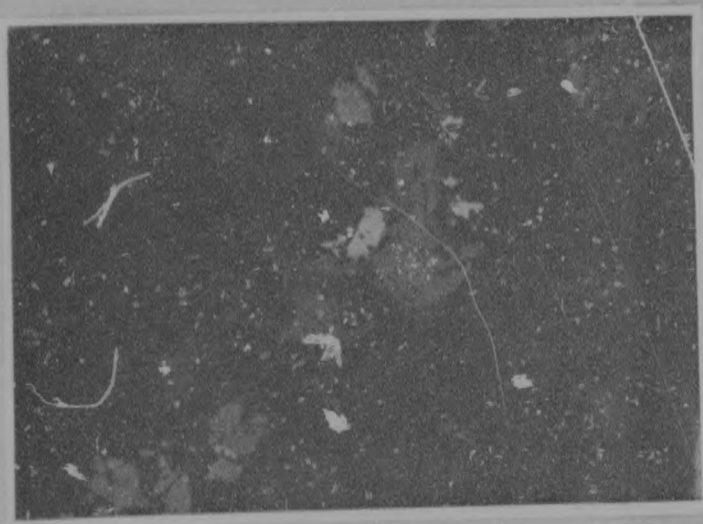


PLATE NO. 23 : Magnetite being replaced by
Arsenopyrite and quartz. Note
tendency to banding in magnetite
diagonally across field.
Polished section. Plane polarised x 250.

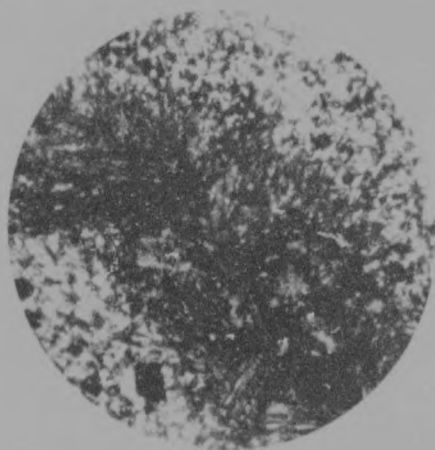


PLATE NO. 16 : Band of felted Amphibolite (Grunerite) needles in Jaspilite: Note opaque mineral centre to band is magnetite; as found in altered siltstones, shales and Jaspilites. X nicols x 52

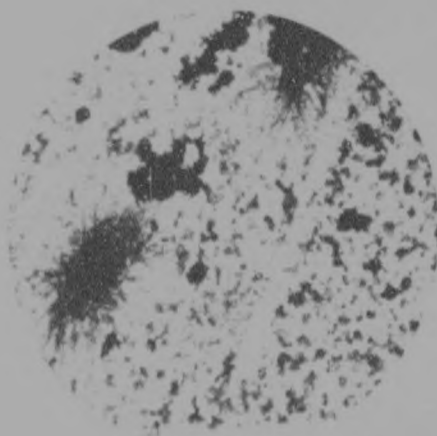


PLATE NO. 17 : Stellate growths of Grunerite around Magnetite grains in Jaspilite. Plate polarised x 62

The arsenic bearing mineral, locally called arsenopyrite, occurs throughout the area of Vubachikwe Mine, associated with economic gold values. It has been found by mining experience that gold values are related to the grain size of the arsenical mineral with which they are associated. Generally, higher gold values are found with fine grained arsenopyrite, although not in all cases. This is not really an uncommon feature in mines of this type: for example, fine grained arsenopyrite is associated with relatively high gold values in the following Rhodesian Mines:-

MARVEL MINE, Filabusi;
 MONARCH MINE, Pinalonga;
 BLANKET MINE, Swanda.

Generally, fine sulphides rather than coarse sulphides are associated with high gold values on the following Rhodesian Mines:-

CONSERVATOR MINE, Hunter's Road;
 GAIKA MINE, Queque;
 VALLEY MINE, West Nicholson;
 EXPRESS MINE, Mashaba.

All these instances have been noted by the writer in the course of work on these mines. It was suspected that the reason for this association on Vubachikwe Mine may lie in the possibility that the coarse and fine grained varieties may represent minerals of varying composition. Thus a series of X-ray analyses was carried out on grains of varying size.

The result . . . / -

The following figures have been used.

<u>TEMPERATURES OF FORMATION:</u>	<u>CENTIGRADES:</u>	<u>EXPERIMENT:</u>
Quartz	600° - 100°	Nicolaysen (1961)
Tourmaline	400° - 450°	Smith (1949)
Pyrrhotite	400° - 600°	Friedman (1959)
Chalcopyrite	350° - 400°	Vokes (1955)

FEATURES OF THE ORE SOLUTION:

There are several notable features of the mineralising solution. It may be stated, from the absence of pyrite and the type of minerals contained in the ore, that, relative to the metallic ion content, the solution was poor in sulphur. With the sole exception of chalcopyrite (which occurs in very small amounts) the minerals found in the ore are of the sulphur poor varieties. Thus the ore contains:-

Arsenopyrite-tellingite, instead of arsenopyrite;

and

Pyrrhotite, instead of Pyrite.

The cause of this lack of sulphur is not known.

A further feature is the uniformity of the ore, and hence ^{of} the solution from which the ore minerals and gangue minerals was crystallised. It is true that

the Vubachikwe . . / -

GANGUE

MINERALS:	Quartz	75%	Estimated proportion	
	Amphibole	10%	"	"
	Carbonate	3%	"	"
	Tourmaline	rare	"	"
	Carbon	rare	"	"
	Garnet	rare	"	"

ORE

MINERALS:	Magnetite	1%	"	"
	Arsenopyrite-			
	Lellingite	7%	"	"
	Pyrrhotite	4%	"	"
	Chalcopyrite	rare	"	"
	Gold	rare	"	"

* * * * * * *

The estimated percentages of minerals comprising the ore are based on visual examinations of washed samples of the mill feed.

Hand specimens of the ore were found to contain a very variable proportion of sulphide minerals, and in highly mineralised specimens, the amount exceeded 50% by volume on visual examination. In the milling of the ore, an average amount of 12% of sulphides is assumed.

An examination . . / -

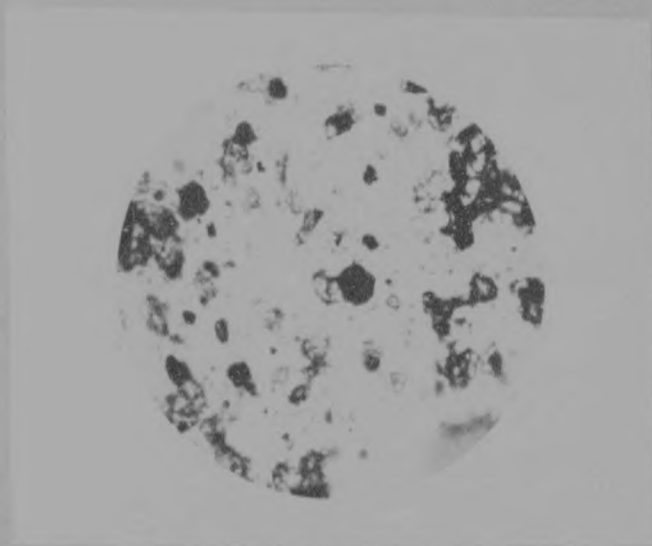


PLATE NO. 18 : Fine subhedra of magnetite disseminated in quartz in jaspilite.
Little or no amphibole is apparent.
Plane polarised x 200.

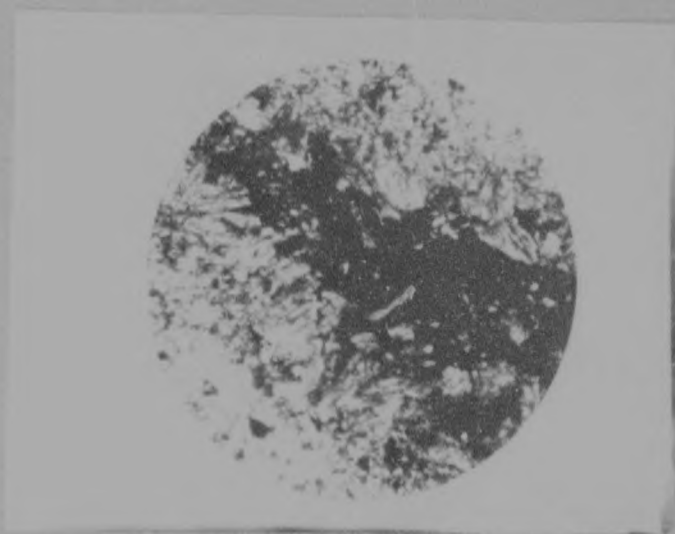


PLATE NO. 19 : Amphiboles growing towards quartz in jaspilite.
Plane polarised x 50

the Vubachikwe property has been mined to no great depth (1,500 ft.). However, from the top of the sulphide zone to the bottom of the mine there is no sign of any mineral zoning. Zoning may become apparent when the mine reaches greater depth.

According to Lundgren's classification of hydrothermal deposits, the mineral association of Vubachikwe Mine places it in the lower range of the hypothermal class. As such, the deposit will probably continue comparatively unchanged for a very considerable distance underground, subject to the existence of suitable structural conditions such as the continued presence of certain host rock and fracture pattern.

ORIGIN OF THE MINERALIZING SOLUTION.

A hydrothermal solution of high mesothermal type impregnated the limbs of the north-west Swanda syncline, and from it orebodies were deposited in favourable localities. These were examined in one particular area (Vubachikwe Mine) and in that limited field were found to contain sulphides of arsenic, iron and copper as well as quartz, tourmaline and native gold. At some later stage the rocks in the district were quite considerably carbonated. There are at least two possible theories for the origin of

this mineralizing . . . / -

The result proved that all the samples were of the same composition. However, the identification of the mineral presented a problem. The latest edition (1961) of the "X-ray Powder Data File" published by the American Society for the Testing of Materials (Pa.) contains no reference to arsenopyrite, previous work on this mineral having been occluded due to unreliability. Standard minerals of known composition were not available, nor were funds or facilities for a detailed chemical analysis. The arsenic-iron-sulphides form a completely isomorphous series between arsenopyrite $\text{Fe As}_2\text{S}_2$ and lollingite Fe As_2 . X-ray results, obtained for the Vubachikwe mineral, were very close to those quoted for reference by Harcourt (1942) as lollingite, and only vaguely resembling those for arsenopyrite. However, since the figures do not correspond exactly and qualitative tests proved the presence of sulphur, the mineral has been accepted as belonging to the isomorphous series arsenopyrite-lollingite, and of composition close to lollingite. (For the X-ray data, see the Section on Mineral Identification by X-ray Diffraction.)

Pyrrhotite:

Pyrrhotite constitutes about 4% of the ore by volume. It is established as being the mineral pyrrhotite (approximating to the composition $\text{Fe}_{1-x}\text{S}_{x+1}$) by analyses using X-ray diffraction. (See section on Mineral Identification by X-ray Diffraction). Results obtained correspond exactly with those obtained for similarly analysed standards. Mine metallurgical

tests . . / -

An examination of a sample of the mill concentrates was carried out under the binocular microscope, and the following proportions of minerals were found:-

MINERAL PROPORTIONS IN MILL CONCENTRATES.

GANGUE	approximately	10%
MAGNETITE	"	3%
ARSENOPYRITE	"	59%
PIRROTITE	"	27%
CHALCOPYRITE	"	less than 1%
GOLD	"	present but rare.

* * * * *

GANGUE MINERALS:

QUARTZ.

In the ore two different types of quartz occur.

Original Quartz:

Considerably more than half the total volume of the ore consists of original quartz, hence the local term 'quartzite' for the ore. This quartz is very fine-grained, and consists of dark grey-blue cherty quartz grains. On fractured surfaces it has a slightly sugary texture. The blue varieties are translucent

in thin flakes . . / -

this mineralising solution:-

a) CUPOLA ORIGIN:

In the north-west Gwanda area, there are two cupolas - or stocks of quartz-monzonite. These are the Insindi and Sizese stocks. It is suggested that these cupolas are evidence of igneous activity in the area, and the mineralising solutions may have emanated from the centres of such activity and have formed orebodies in suitable places. It has been shown in the section on Structure that the fracture pattern followed by the solution was probably formed by the "second deformation". Therefore, the igneous activity must have followed the "second deformation". However, it has not been possible to determine the relative age of the quartz-monzonite cupolas. Their field relationships with the country greenstones are not at all clear: No clear contacts have been seen, and although no "second deformation" axial plane cleavage has been found in the stocks, it cannot definitely be stated that the cupolas are post second deformation.

Because no orebodies have been found directly associated with the two cupolas, it does not necessarily mean that no mineralising solution was with them. The cupolas are, however, evidence of quartz-monzonite igneous activity, and such activity is associated in many parts of the world with gold mineralisation.*

b) LATERAL SECRETION:

The possibility . . . / -

* Simons, 1937.

tests have revealed no nickel in the ore, and no pentlandite has been observed intergrown with the pyrrhotite. Pyrrhotite occurs in irregular grains and veinlets throughout the ore from submicroscopic up to four inches across, and is also noted in the slaty bands where it occurs in grains and as smears on partings of the slate. It is of the usual typical metallic brown colour when fresh, and tarnishes rapidly on oxidation. Tarnished surfaces sometimes show a brilliant peacock iridescence not unlike bornite, but more brassy-coloured. Euhedral crystals of pyrrhotite were not observed. In limited localities, associated with introduced quartz, concentrations of pyrrhotite as large as four inches across have been seen. Compared with the magnetite in the ore, the pyrrhotite is only weakly magnetic.

Under the reflecting microscope, the pyrrhotite appears a pinkish cream colour and contrasts sharply with the bright white of the arsenopyrite. It is also comparatively soft, being easily scratched with a steel needle. In polarised light, the pyrrhotite is quite discernible as anisotropic changing from a dark grey-brown to a paler shade of the same colour.

The pyrrhotite began to crystallise with the last fraction of the arsenopyrite, and is frequently associated with it in completely mutual relationship (see Plate No. 30): pyrrhotite replaced gangue to a fairly large extent and, to a limited extent, corroded and replaced arsenopyrite. Evidence for the continued crystallisation of pyrrhotite later than the arsenopyrite is seen in the form of veinlets of

pyrrhotite . . / -

in thin flakes. There is a tendency for the original quartz to occur as bands in the rock - apparently conforming to the original bands of the banded jaspilite. However, so much of the original texture has been obliterated by mineralization, that this banding is no longer a well marked feature. In thin sections, the quartz is seen to consist of a pavement-like mosaic of moderately interlocking quartz grains all of well sorted similar size. Occasionally very small grains of magnetite are present. There is a tendency for the pavement mosaic, as a whole, to follow a banded structure. This is sometimes accentuated by the regular orientation of amphibole and chlorite crystals which are occasionally present. This cherty quartz is very hard and, with its "reinforcing" of amphibole needles, has become a particularly tough rock. The dark-greyish colour may be due to a small amount of carbon contained in the rock. (See Plates Nos. 10, 11, 17 and 18.)

Introduced quartz:

All the quartz observed of this nature is of "vein quartz" type. The veins are seen filling all manner of fractures in the rocks, from comparatively large fault-fillings, such as the two foot wide vein on 11 Level footwall drive, to tiny microscopic veins healing cracks in the massive ore. Throughout the Yuba-Long-John sections of the mine it is a pale grey glassy looking quartz, showing pronounced shattering. It often contains wispy growths of tremolite. The interior of the large veins (i.e. those greater than four inches thick) consist of comparatively pure quartz,

while the margins . . . / -

The possibility that the orebodies were formed by lateral secretion must be considered. The term is used reservedly and the agency for secretion was not meteoric waters as propounded in the early theories of Baubree (1887) but rather envisaged along the following lines : - *

All the elements found in the Vubachikwe orebodies (except Pb and Au) are present to a greater or lesser extent, some as minor constituents in Rhodesian greenstones and granites.** Sulphur has been found in amounts up to 0.10%, and arsenic 0.05%. Unfortunately, no rock analyses are available for the area under discussion, but no logical reason can be seen for the area to be an exception to Rhodesian gold-belt rocks. Thus it is suggested that all the required elements to form orebodies were spread through the regional rocks. Under metamorphic conditions, probably at considerable depth, the marginal portions of the greenstones were subjected to granitizing conditions. Processes of granitization were already advanced in the rocks in which the greenstone complexes lay, thus the ore minerals together with carbonates and silica were mobilised into solutions within the rock. During the metamorphism, the solutions migrated into fractures and veins, and were forced away from areas of greatest compression.*** Hence, they came to be hydrothermal solutions. The minerals

from these . . . / -

*Ref. Bateman, Chapter 1.

**Holding, 1936.

***MacGregor, 1951.

pyrrhotite formed in arsenopyrite and original quartz, and also as partial rims to arsenopyrite crystals, as well as filling interstitial cracks and occurring as graphic growths in quartz. The pyrrhotite was partially replaced by later quartz in places. See Plates Nos. 32, 33, 34, 35 and 36.

CHALCOPYRITE:

A very small amount of chalcopryite was found in the ore. At the end of the stage of crystallisation of pyrrhotite, chalcopryite crystallised, partly contemporaneously with the pyrrhotite, and partly replacing pyrrhotite. (See Plate No. 37.) The chalcopryite is a yellow soft sulphide, positively identified by Xray diffraction analyses. (Ref. section on the Identification of Certain Minerals by Xray Diffraction.) The largest grains seen are associated with pyrrhotite and are up to 12 millimetres across: however, usually the chalcopryite forms a very insignificant portion of the ore.

Under the reflecting microscope, the chalcopryite is seen to be a yellow colour which, in very finely divided grains, can be mistaken for gold. The larger grains are easily distinguished by their brittleness, compared with the malleability shown by the gold. Also when the two are seen together, the gold has a deeper yellow colour and a much higher reflectivity. The chalcopryite does not exhibit pleochroism. Under reflected polarised light it is slightly anisotropic, varying in shade of yellow-brown on rotation of the

microscopic stage. / -

while the margins may contain considerable amounts of sulphides. There is a very close association between this "clean" quartz and all the "visible" native gold seen on the mine.

In the Magano section the quartz has a dense white milky colour. This may be due to a faster rate of cooling, forming smaller crystals than the quartz of the northern parts of the property. However, it is thought this effect may be due to a carbonate content in the quartz. A gentle gradation from the highly vitreous and glassy quartz in the north to the milky white variety in the south, is believed to exist. There is insufficient data available on the 2,500 foot space between the Magano and Long John sections to prove this contention, although surface indications are that generally there is an increase in the carbonate proportion of the rock southwards. (See plates Nos. 12 and 45).

CARBONATES:

Whilst a portion of the undifferentiated sediments was probably calcareous, the major portion of the carbonates are believed to have arisen through carbonitisation of the country rocks. As such, most of the carbonates have been introduced into the surrounding greenstones and relatively little occurs in the actual ore. The ore is, however, transversed by many carbonate veinlets. They fill cracks and small fractures which cut across all the other minerals and structures. It is considered that a large proportion of these veinlets (up to two

inches thick) . . / -

from these solutions would be replaced in chemically and physically suitable environs. The general impression of the Canada Area is very similar to the Yellowknife Goldfield of the North-West territories of Canada: it has been shown that the economic deposits there were formed by this type of lateral secretion,* and it is believed that this could have been the mode of origin of the mineralising solution in the north-west Canada area.

*Payle, 1959.

microscope stage. The chalcopyrite is rarely seen associated with arsenopyrite, being usually separated from it by gangue or pyrrhotite. Occasionally it is enclosed in pyrrhotite (see Plates Nos. 37 and 38), but more often fills up little spaces at the ends of pyrrhotite veinlets (see Plates Nos. 32, 33 and 38), or occurs between grains of pyrrhotite. Thus since it is found at extremities of the pyrrhotite, it probably was introduced at a later stage of pyrrhotite crystallisation and, as the chalcopyrite crystallised, it replaced small amounts of pyrrhotite.

GOLD:

Gold is the mineral of economic importance in this orebody. Rich specimens containing visible gold are rarely found on Vubachikwe Mine. In order to examine the gold more closely, samples of concentrate obtained from the spillage James table in the mill were panned, and the product, enriched in gold, was mounted and prepared in polished sections. Coarse gold is known to occur on the mine since nuggets of several millimetres in size are recovered from the ball-mill periodically. The only available coarse gold actually seen in situ was taken from the Magano Section. (Plate No. 45.)

Polished sections of the gold under the reflecting microscope show it to be a very rich yellow, and under polarised reflected light it shows a distinct but weak anisotropism. Although persistent buffing of the section fails to remove this anomaly, it is a polishing effect. No crystals of gold have been seen, all the grains being irregular in shape: doubtless, some of the

grains . . / -

inches thick) was deposited by percolating ground waters, which dissolved carbonates from the greenstone, and redeposited them in cracks in the orebodies. However, since some of these veinlets contain carbonates of highly crystalline nature, it is thought that these may have been precipitated from hydrothermal solution at the end of the mineralizing phase. No analytical work has been done on the carbonates, but hard specimen examination suggests that the bulk of the mineral is calcite, while some very brown and well crystallised portions are thought to be siderite. The ore of the Magano Section contains a very much greater proportion of carbonates than the ore of the main sections. This is not surprising since the local rocks around the Magano Mine are much more calcareous than those further north.

It is interesting to note that although the ore contains carbonates, these are present in insufficient quantity - or insoluble form - to preserve the protective alkalinity necessary for the cyanidation process used by the mine; and considerable lime has to be added to the mill. (See Plate No. 13).

AMPHIBOLES:

The clay minerals, iron minerals and the fine grained quartz originally present in the banded iron-stones probably reacted under metamorphic conditions to form iron-rich amphiboles. Some of these amphiboles have subsequently been altered to chlorites. Exceptionally, the amphiboles form as much as 40% of the rock

by volume; . . / -



Arsenopyrite band in ore. Note
subhedral-subhedral crystals in quartz
PLATE NO. 24 : and magnetite, and penetration crystals.
Polished section.
Plane polarised x 82.5.



Arsenopyrite. Subhedral crystal shattered
and being replaced by quartz and tourma-
line in quartz vein. Tourmaline is dark
PLATE NO. 25 : gray mostly at bottom of white arseno-
pyrite crystal. Polished section.
Plane polarised x 82.5.



PLATE NO. 20 : Tourmaline needles in quartz,
associated with arsenopyrite
in quartz veins.

Plane polarised x 50.

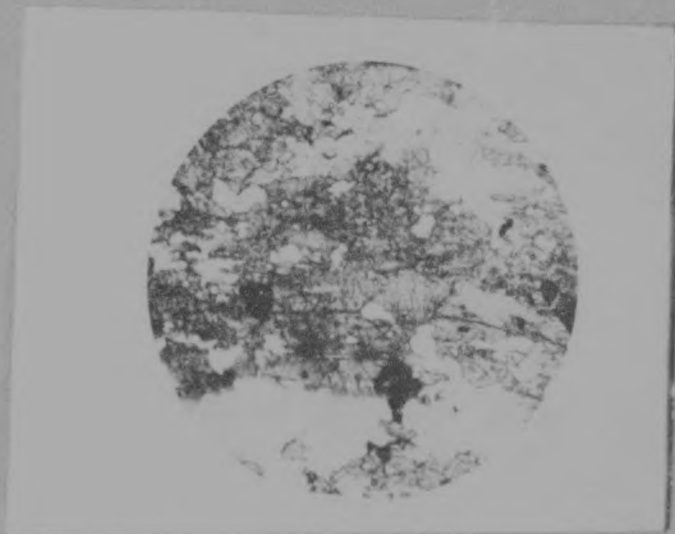


PLATE NO. 21 : Tourmaline schist. Tourmaline
impregnated in country greenschist
in close proximity to tourmaline
quartz vein.

Plane polarised x 50.

by volume; but usually only about 10%. The amphiboles were identified by Phaup and are discussed in his unpublished report dated 2.3.1935. Grunerite is the commonest amphibole, and occurs as finely divided, thin, felted crystals. These crystals often radiate from magnetite bands into quartz bands in the jaspilite. Where the original iron content was low, hornblende has formed rather than grunerite; and in the immediate vicinity of quartz veins, tremolite occurs, occasionally in crystals over an inch long. (See also the description of amphiboles in section on Jaspilite, under Local Geology.) See Plates Nos. 6, 7, 8, 14, 15, 16, 17 and 19.

TOURMALINE:

The only tourmaline seen on the mine occurs associated with very coarse arsenopyrite in and near quartz veins. It is sometimes seen to penetrate the rock through which the quartz vein passes, generally with crystals orientated at about right angles to the plane of the vein. In some cases a band consisting almost entirely of tourmaline is formed on the edge of quartz veins. The size of such occurrences averages about one inch of tourmaline on each side enclosing about three inches of quartz. Typically, the tourmaline forms masses of combined crystals, usually about 0.5 millimetres thick, and up to 15 millimetres long. They are all intense black in colour. Microscopically, the crystals show very typical cross fracturing along their length. They are coloured a very deep brown and are very pleochroic. The pleochroic formula is C - transparent pale buff; E - dark greenish brown. From the association with arsenopyrite, the tourmaline is considered to have crystallised at an early stage in the mineralization.

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CHAPTER EIGHT
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CONCLUSION
ON
STRUCTURE
AND
MINERALIZATION
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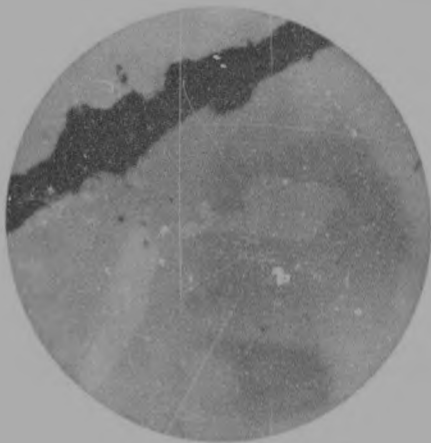


PLATE NO. 37 : Chalcopyrite completely enclosed in
pyrrhotite and also as a rim to small
quartz vein filling crack in pyrrhotite.
Probably all particles are connected by
veinlets of chalcopyrite.
Polished section. Plane polarized x 1200.

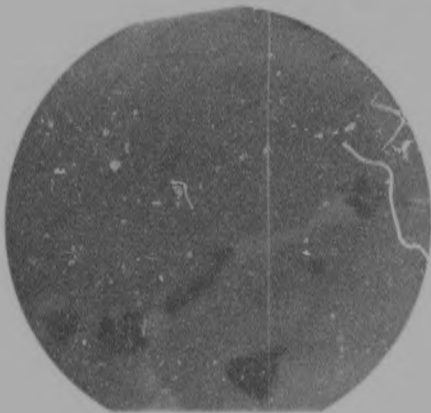


PLATE NO. 38 : Chalcopyrite as residual filling of
crack in pyrrhotite partially filled
with quartz and filled with chalco-
pyrite. Polished section.
Plane polarized x 1200

<u>Year:</u>	<u>HILL</u>			<u>CLAYVILLE</u>			
	<u>Tons:</u>	<u>Gr. Au:</u>	<u>dwt/ton:</u>	<u>Tons:</u>	<u>Slimes:</u>	<u>Gr. Au:</u>	<u>dwt/ton:</u>
191	7	46	7	-	-	-	-
1920	10,000	2,656	5.31	-	-	-	-
1924	20,150	5,777	5.73	-	-	-	-
1925	10,300	2,164	3.97	6,000	-	with mill	-
1926	8,500	1,894	4.46	3,500	-	with mill	-
1927	-	1	-	-	-	-	-
1932	-	5	-	1,500	-	75	1.0
1933	4,484	710	3.17	2,907	-	with mill	-
1934	11,450	2,315	3.87	4,890	480	-	-
1935	10,193	592	1.16	22,694	-	1,072	0.95
1936	12,100	753	1.74	24,000	100	1,591	1.32
1937	9,500	868	1.83	30,000	1,900	1,643	0.71
1938	9,735	2,085	4.28	34,350	2,550	with mill	-
1939	4,550	2,540	11.16	17,700	1,900	with mill	-
1940	16,250	3,078	3.79	2,700	2,600	with mill	-
1941	18,800	4,496	4.79	17,700	-	591	0.67
1942	26,000	5,574	4.29	9,497	-	3.2	0.64
1943	33,300	6,273	3.77	-	-	-	-
1944	32,600	6174.5	4.14	-	-	-	-

CHAPTER EIGHT

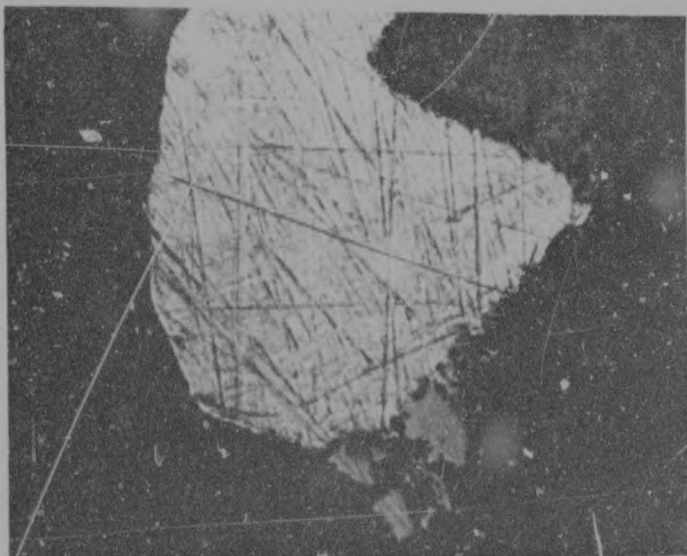
CONCLUSION ON STRUCTURE AND MINERALIZATION.

The present structure of Vubachikwe Mine arose through a complex succession of geological processes: these processes have been sub-divided into a series of three tectonic disturbances accompanied by considerable metamorphism. The introduction of economic minerals is associated with the second tectonic disturbance which resulted in the formation of a regional synform. Vubachikwe Mine lies in the north limb of this synform; the localisation there of orebodies is dependent on a continuation of the nature of the rocks being replaced, and on structural features dependent on - or parasitic to - the regional synform. These features are mainly a system of plunging folds along the limbs of the synform, some folds being quite intense, others very gentle.

The localisation of minerals in a sequence of plunging folds is a common occurrence in Precambrian mining areas of the world.

The mineral association seen in the ores of Vubachikwe Mine is considered ample evidence of its high temperature mesothermal conditions of deposition. Although it is possible that zoning may become apparent when a greater vertical section is visible, the deposit has so far been very uniform in the sulphide zone. The mineral solutions are particularly characterized by their having been of low sulphur content, the mineral ensemble showing a distinct paucity of that element.

Thus . . . / -



Gold-in groundmass of bakelite - mounted
polished concentrate. Peculiar striation
effect was originally believed to be silver
exsolving. Subsequently found to be polish-
ing effect only. Plane polarized polished
section x 200.



Gold showing inclusion of pyrite.
Plane polarized section x 1000

<u>BILL</u>					<u>CYANIDE</u>			
<u>Year:</u>	<u>Tons:</u>	<u>Oz. Au:</u>	<u>dwt/ton:</u>	<u>Panda:</u>	<u>Slimes:</u>	<u>Oz. Au:</u>	<u>dwt/ton:</u>	
1945	36,000	7,087	3.93	-	-	-	-	-
1946	35,800	7,121	3.98	-	-	-	-	-
1947	33,900	7,397	4.36	-	-	-	-	-
1948	31,300	6,745	4.31	-	-	-	-	-
1949	34,700	5,637	3.81	-	-	-	-	-
1950	35,260	7,128	4.04	-	20,380	266	0.26	-
1951	33,000	5,921	3.39	2,900	-	-	-	-
1952	9,454	1,311	2.77	-	30,645	860	0.51	-
1953	10,538	906	1.77	-	8,361	394	0.94	-
1954	55,122	8,681	3.13	-	-	-	-	-
1955	63,374	8,822	2.56	-	-	-	-	-
1956	73,976	11,489	3.10	-	-	-	-	-
1957	75,790	11,835	3.12	-	-	-	-	-
1958	81,830	15,988	3.90	-	-	-	-	-
1959	77,068	19,906	3.16	-	-	-	-	-
1960	5,424	17,805	4.66	-	-	-	-	-
TOTALS :	1,008,850	199,223	3.85					

Thus this deposit may be considered as a high temperature mesothermal replacement, or impregnation along a system of pre-existing fractures with a very minor amount of open-space filling, most of the sulphides being of wall rock replacement type.

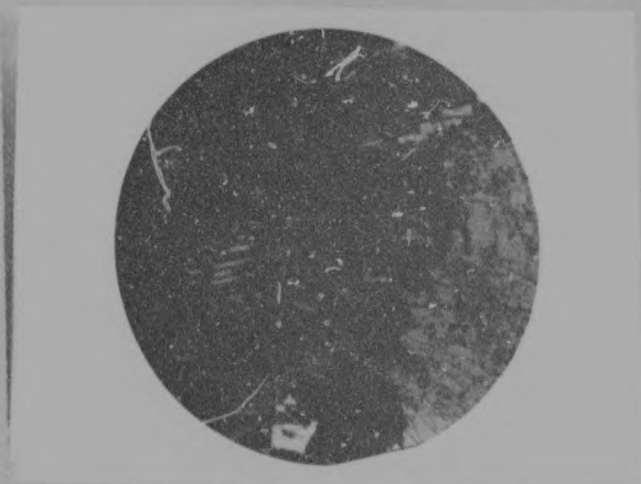


PLATE NO. 41

Gold in groundmass of Amphibole and
quartz. Note linear structure of gold
to conform with grange crystal structure.
See Plate 42. Plane polarised x 62

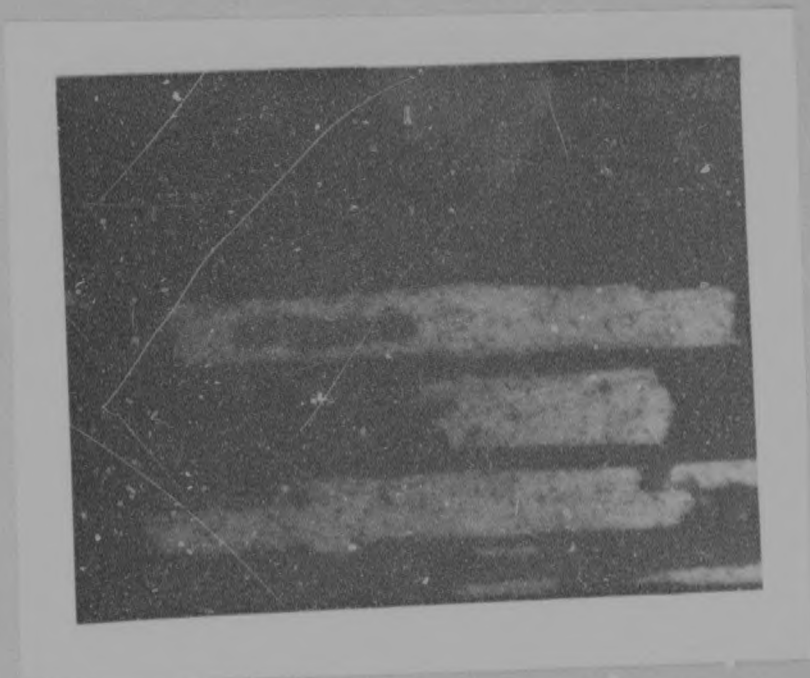


PLATE NO. 42 : Gold - enlarged from Plate No. 41.
Note Pyrrhotite inclusions in gold.
Plane polarised polished section x 5500

W I L L				O T A S I D E			
<u>Year:</u>	<u>Tons:</u>	<u>Gr. Bu:</u>	<u>Div/ton:</u>	<u>Grades:</u>	<u>Slimes:</u>	<u>Gr. Bu:</u>	<u>Div/ton:</u>
1945	36,000	7,087	3.95	-	-	-	-
1946	55,800	7,121	3.96	-	-	-	-
1947	53,900	7,597	4.36	-	-	-	-
1948	51,500	6,745	4.31	-	-	-	-
1949	54,700	6,637	3.81	-	20,880	266	0.26
1950	55,260	7,128	4.04	-	-	-	-
1951	55,000	5,521	3.39	2,900	50,645	660	0.52
1952	9,454	1,311	2.77	-	8,561	394	0.34
1953	10,538	506	1.77	-	-	-	-
1954	55,122	8,681	3.15	-	-	-	-
1955	68,374	8,822	2.56	-	-	-	-
1956	73,978	11,489	3.10	-	-	-	-
1957	75,790	11,635	3.12	-	-	-	-
1958	81,830	15,988	3.90	-	-	-	-
1959	77,068	19,906	3.16	-	-	-	-
1960	76,424	17,805	4.66	-	-	-	-
TOTAL: 1,106,650				193,223	3.83		

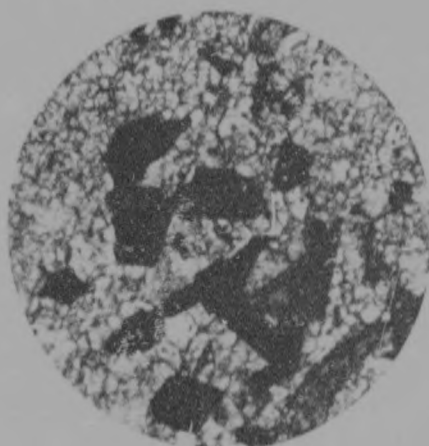


PLATE NO. 26 : Arsenopyrite in Jaspilite ore. Note
opaque arsenopyrite in matrix section.
Thin section.
Plane polarized x 50



PLATE NO. 27 : Arsenopyrite. Note aggregate and discreet
crystals in ore. Polarized x 300.
Polished section.

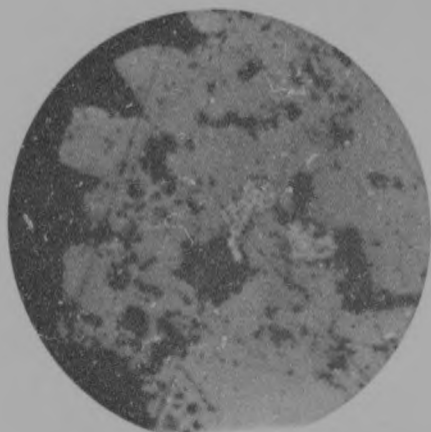
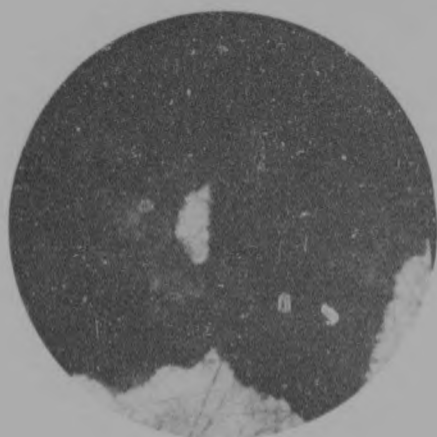


PLATE NO. 43 : Gold in fine grained aggregate arsenopyrite. Black sections are gangue.
Plane polarised polished section x 330.



Gold in reacted arsenopyrite -
(probably converted to magnetite)
PLATE NO. 44 : in reacted concentrate mounted in
bakelite. Plane polarised polished
section x 1200.

In addition, the following tonnages have been declared separately, having been worked on tribute:-

VUBACHIKWE MINE:

1940 - 1957 47,542 tons 8,039 ozs. 3.39 dwt/ton

NASATO MINE:

1934 - 1941 59,915 tons 10,564 ozs. 2.37 dwt/ton

Thus, to date, a minimum of 1,231,000 tons have been mined from the property; and 241,000 ounces of gold produced of an approximate grade of 3.9 dwtz/ton.

* * * * *

H I S T O R Y

It is said that ancient workings led to the rediscovery of gold in the north-west Gwanda gold belt about 1890. In the early 1890's a host of "pioneer workings" were started. Prior to 1894, practically nothing is known of the Vubachikwe Mine area: in 1894 the area was acquired by the Rice Hamilton Exploration Syndicate. That this mining venture failed miserably seems due to inefficiency and bad management. By 1911 the syndicate was bankrupt and the property was taken over by the firm Rhodesia Limited. The Vubachikwe string of claims lay dormant from 1912, except for a short period in 1916, and passed to the London and Rhodesia Mining and Lands Company Limited, (LonRho), in 1922. They tributed out portions of the area and

the main mines . . / -

CHAPTER NINE

THE IDENTIFICATION
OF CERTAIN MINERALS
BY X-RAY DIFFRACTION

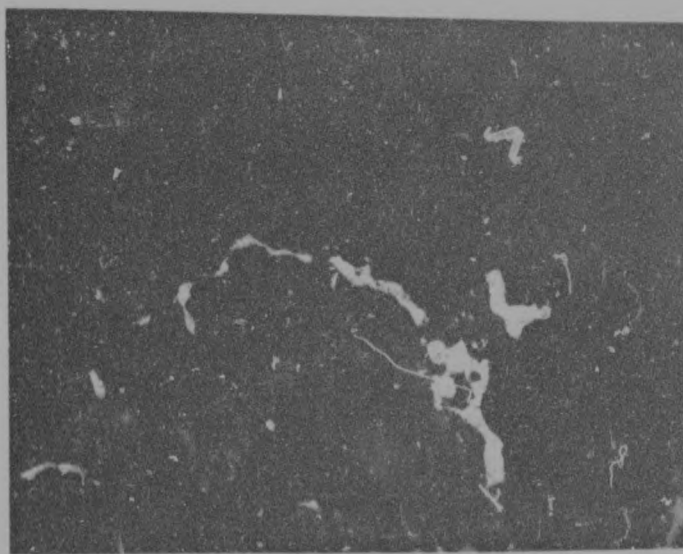
* * * * *

* VIVIANITE-VAUXITE *

* CHALCOPYRITE *

* PIRROTITE *

* ARSENOPYRITE-LOLLINGITE *



Gold as veinlet or interstitial filling
PLATE NO. 45 : in secondary quartz. Same polished
polished section x 250.

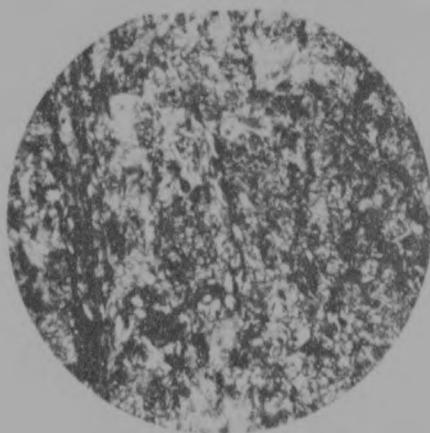


PLATE NO. 46 : Calc schist composed of talc,
chlorite and carbonate.
x nicols x 50.

the main mines produced intermittently. From 1933 to 1938 London worked the main sections of Vubachikwe at a loss. In 1938 the Rosemont General Mining Company, a London subsidiary, acquired the mine area and tributed it to Forbes & Thompson (Canada) Limited, who purchased it in April, 1939.

Since then, through careful management and practical mining and milling, the mine has proved a profitable proposition. The present tonnage is about 6,000 tons per month (1961).

GENERAL:

Although a comparatively small mine, Vubachikwe is run on modern methods employing a considerable degree of mechanization, including the use of electric locomotive haulage on the 10th level. Normal mining practice is employed, and broken ore is brought to the surface with as little handling as possible. The broken ore all passes through a 12 inch grizzly before being hauled to the surface. The main shaft is the Arthur Thompson vertical shaft (See Plate No. 4) and is well equipped to the bottom of the mine.

SORTING:

Although sorting experiments carried out on the property have been successful, serious sorting of ore has not been carried out, since the mill capacity is greater than the tonnage mined.



PLATE NO. 47 : Dolomite - Chill zone.
Plane polarized x 12.

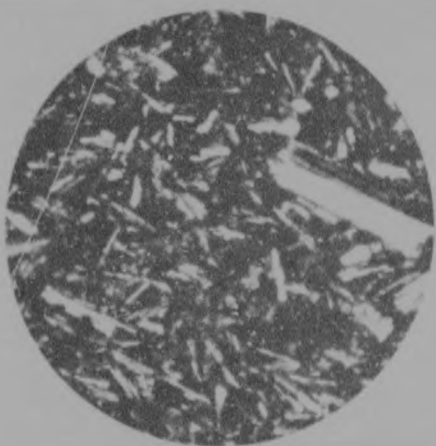


PLATE NO. 48 : Dolomite - Chill zone.
Plane polarized x nicols x 62.

T H E M I L L

A study of the diagram of Subochikwa Mill (Page No. 127) shows nothing out of the ordinary for a gold reduction plant. Ore is crushed, classified, tabled and floated, and the concentrate is subjected to a simple cyanidation.

Within the limits of economics, and compared with similar mills of other Rhodesian gold mines, and the refractory nature of the ore, this mill is considered efficient. Over 80% of the total gold content of the ore is recovered. Gold is the only product from the mine. The only other possible by-product, arsenic, is not produced commercially. The small amount of silver in the bullion is insignificant.

Two features of the plant are noteworthy:-

1) FINE GRINDING:

Because the gold is so finely divided, an extreme degree of grinding is necessary to free it for cyanidation. Thus, concentrates are reground to a fineness of 94% - 325 mesh.

2) GALLAGHER TABLES:

This is not a metallurgical treatise, but a brief description of these ingenious contrivances is of interest. These tables were developed (and patented) by the Mine's Chief Engineer, Gallagher. The table consists (in essence) . . . / -

essence) of a short broad belt of corduroy tilted at an adjustable angle and rotating very slowly. The top surface of the belt is evenly supported to form a table top, functioning exactly as a normal corduroy or blanket strake. The pulp is fed evenly over the high edge of the table top and runs down over the surface leaving a deposit of concentrate in the riffles. The rotation at right angles to the flow of the pulp ensures an automatic clearing of the concentrates. The top of the table is constantly being replaced, and the concentrate carried over the edge to be removed by water sprays underneath. A gentle jiggling vibration can be applied if necessary. Approximate dimensions of the table are 6 ft. x 4 ft., the diameter of the rollers at each end being about 12 inches, and the speed of the rotation about 12 inches/minute.

The obvious advantage of such a machine is its large capacity for concentrate extraction. It is particularly useful when applied to ores containing a high percentage of heavy material.

FLOWCHART OF VORACHIK-MINE REDUCTION PLANT (VIRG)

... / -

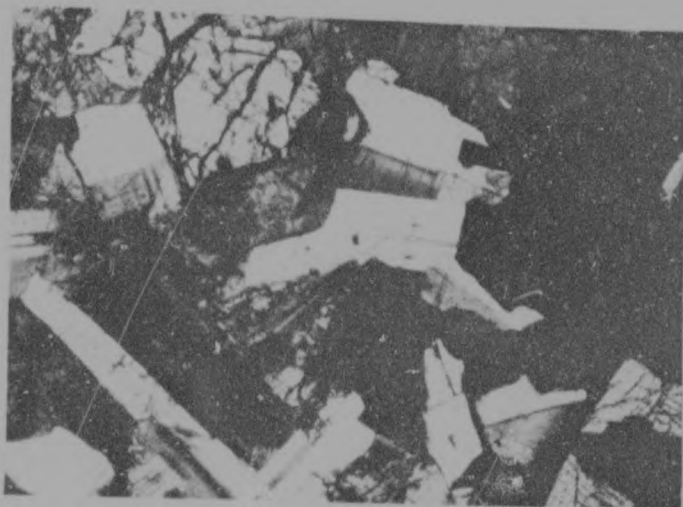
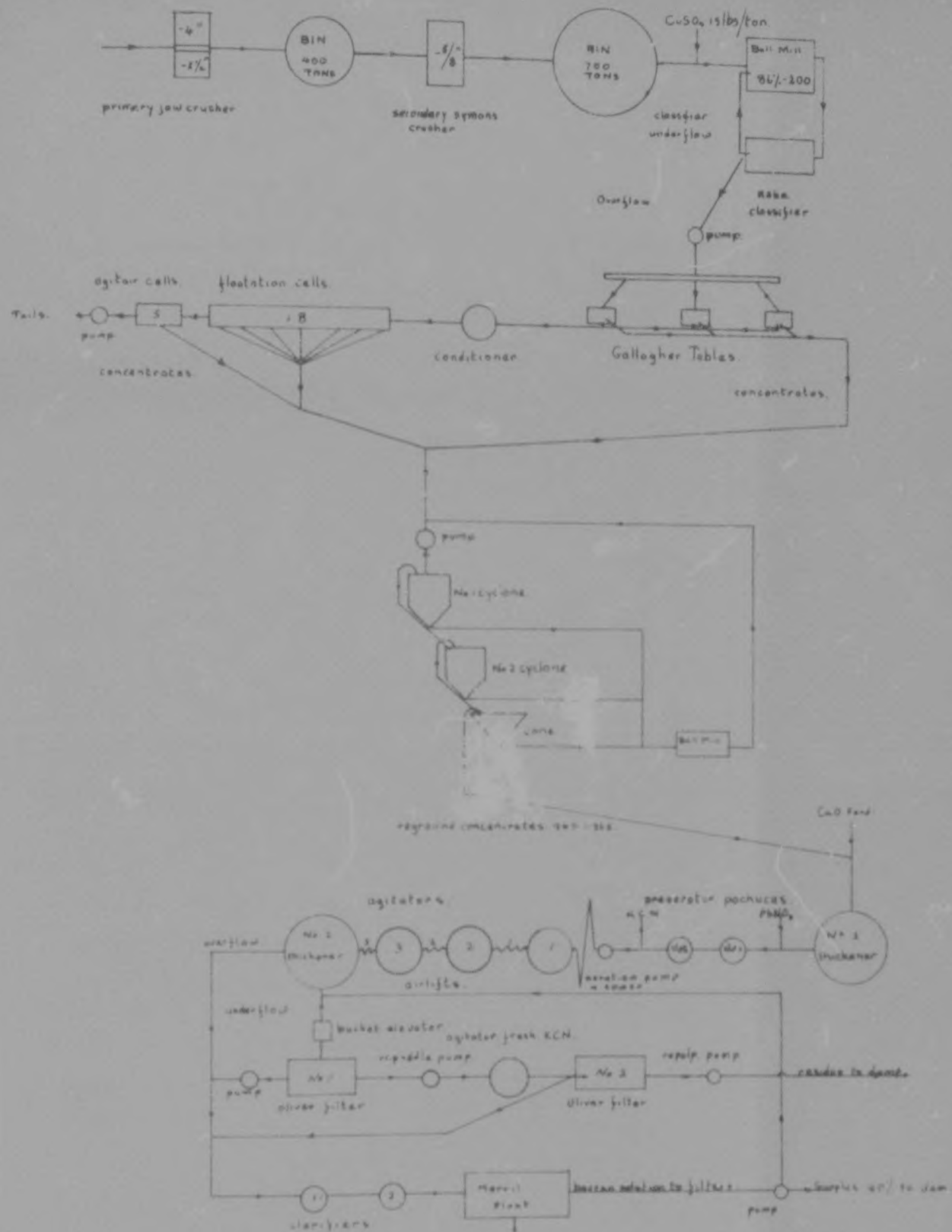


PLATE NO. 49 : Dolerite. Coarse grained from centre
of sill.

1 nicols x 50.



VUBACHIKWE MINE REDUCTION PLANT

May 1961.

20/1/61

B I B L I O G R A P H Y

Canada) and also did not remember the data file figures - but were very close to the Vubchikwe sample.

"d" VALUES:

<u>VUBCHIKWE:</u>	<u>SEDBURY:</u>	<u>DATA FILE:</u>
2.62 (66)	2.61 (66)	2.59 (100)
2.05 (100)	2.05 (100)	2.98 (90)
1.66 (53)	1.62 (60)	2.67 (30)

The data file does not quote the radiation or filter used and this is probably where the discrepancy arises.

ARSENOPYRITE-LÖLLINGITE:

Since gold is associated with a fine grained variety of the arsenic sulphide rather than with the coarse, on Vubchikwe Mine it was suspected that there may be a chemical as well as physical variation in the mineral. A series of analyses on various samples was done using the X-ray powder diffraction method. The results were compared with Marcourt's (1962) figures, since arsenopyrite is excluded from the "X-ray Powder Data File", 1961. From the results obtained, it would appear that the various samples (described below) are all approximately the same and belong to the isomorphous series, arsenopyrite-löllingite, being somewhat sulphur poor and hence

nearer . . . / -

P O V E R

For many years, Subochine Mine has been
linked to the electricity grid of the Electricity
Supply Commission of Southern Rhodesia. Power
is supplied from the station at Gwelo.

.. .. .
.. .. .

F I N

nearer to the lollingite end member of the isomorphous series.

"d" VALUES:

(Barcourt)							
<u>INTENSITY//</u>	<u>ROLLINGITE//</u>	<u>H2</u>	<u>H11</u>	<u>H12</u>	<u>H13</u>	<u>H15</u>	<u>H12</u>
100	2.45	2.38	2.39	2.38	2.38	2.41	2.37
50	2.67	2.65	2.64	2.61	2.63	2.67	2.63
20	1.81	1.79	1.80	1.79	1.80	1.81	1.79

The "d" values for arsenopyrite are considerably different as the line intensities are so different.

"d" VALUES (Barcourt 1942)

ARSENOPYRITE:

INTENSITY:	"d"
100	2.65
100	2.42
20	1.81

The samples listed above are briefly described:-

- H2. Extremely fine arsenic sulphide assaying 10dwts Au/ton.
- H11. Fine 0.5 millimetre aggregate arsenic sulphide - low Au content.
- H12. Very coarse 0.5 millimetre crystal arsenopyrite. Au less than 0.5 dwt/ton.
- H15. Very fine aggregate arsenopyrite, approx. 60dwts Au/ton.
- H15. Extremely coarse arsenopyrite crystals 15 millimetres, Au less than 2 dwt/ton.
- H12. Coarse arsenopyrite associated with tourmaline gold, less than 0.2 dwt/ton.

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PLATE NO. 32 : This veinlet of Pyrrhotite and
Chalcopyrite on crystal of
Arsenopyrite in quartz matrix.
See Plate No. 33.
Polished section.

Plane polarised x 250.

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- " Log of Borehole No. 16. 26. 4. 60.
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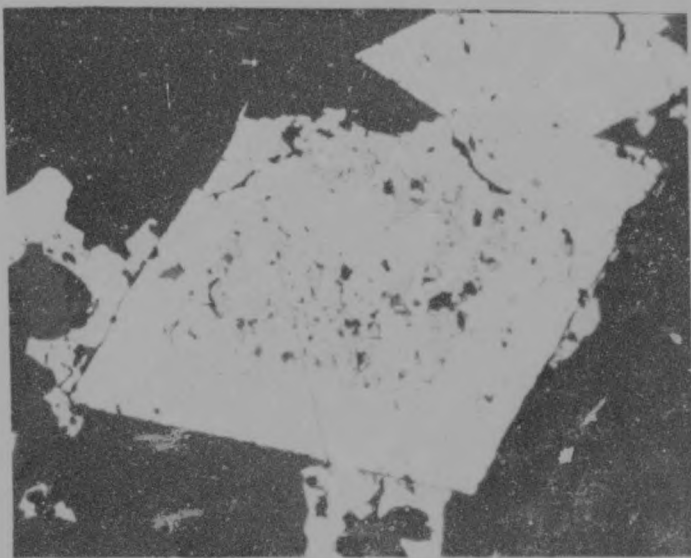


PLATE NO. 30 : Arsenopyrite crystal showing mutual intergrowth and partial replacement by Pyrrhotite. Polished Section.

Plane polarised x 300

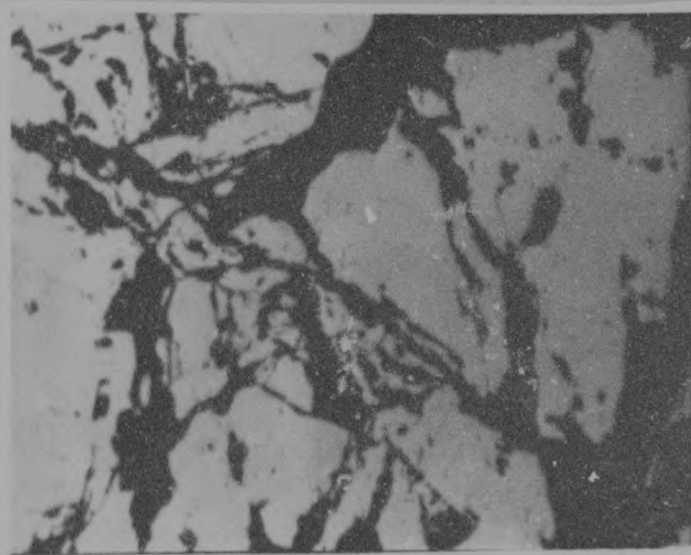


PLATE NO. 31 : Pyrrhotite - shattered and partially replaced by quartz. Note if pieces are "jigsaw"-fitted, there are gaps indicating partial replacement. Polished Section. Plane polarised x 72.5.

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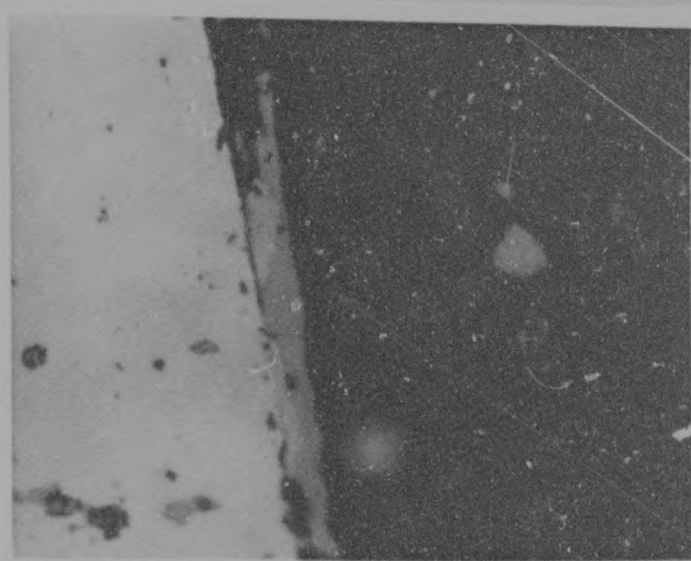


PLATE NO. 33 : Arsenopyrite with fine veinlet of
pyrrhotite and chalcopyrite in
quartz. Polished section.
Plane polarised x 1200

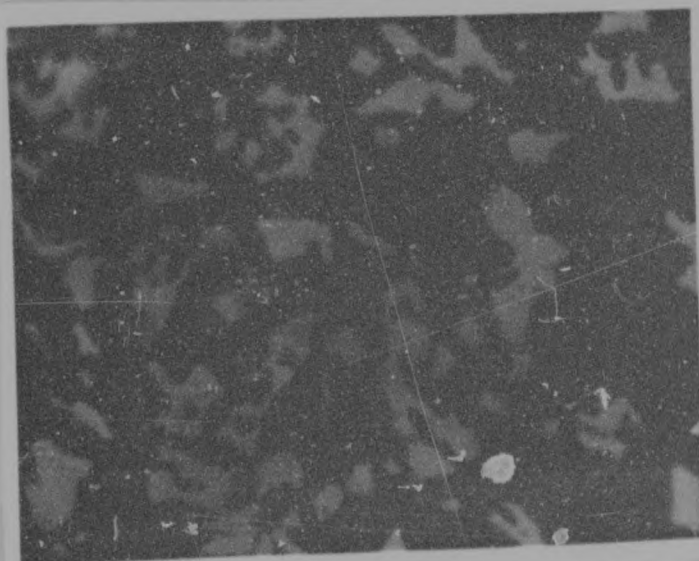


PLATE NO. 34 : Graphic growth of pyrrhotite in
quartz. Polished section.
Plane polarised x 1200.

CHAPTER TEN

OUTPUTS

*

HISTORY

*

GENERAL

*

SETTING

*

T H M M I L L

1. VIBO GRINDING
2. BALLMILL TABLE

*

FLOWCHART OF
VIBOGRINDING MILL
REDUCTION PLANT (MILL)

*

POWER

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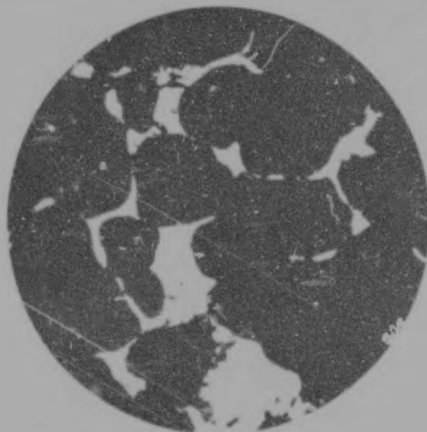


Pyrrhotite being replaced by quartz.

PLATE NO. 35 :

Polished Section.

Plane polarised x 250.



Pyrrhotite as interstitial filling
between quartz grains - probably
original quartz in the Jaspilite.
Polished Section.

PLATE NO. 36 :

Plane polarised x 100.

CHAPTER TEN

OUTPUTS

The Kibichikwe mine has had an erratic history, as is shown in the following output tables.

No output figures are available before 1905.

1905 - 1908; 85,125 tons were milled, recovering 29,212 ounces of gold at an average grade of 6.86 dwts/ton.

OUTPUT TABLES . . . -

CHAPTER NINE

THE IDENTIFICATION OF CERTAIN
MINERALS BY XRAY DIFFRACTION.

Since the latest (1961) Phillips Xray equipment was available, it was decided to use Xray Diffraction for positive identification of certain minerals occurring in the ore.

VIVIANITE-VAUXITE:

This mineral, described in the section on Mineralogy, occurs commonly in the oxide zone in the Baguio section. Since comparatively large specimens were available, it was analysed using the diffraction goniometer with the automatic recording device. The graph obtained showed a substance of almost amorphous nature, with only three very broad, low amplitude peaks being registered. Due to the lack of definition of the peaks, accurate readings were not obtained. The "d" values and intensity are recorded here together with the values quoted in the "Xray Powder Data File" (1961) for delvauxite, the mineral to which the figures most closely correspond.

"d" VALUES:EXAMINED MINERAL: INTENSITY: DATA FILE: INTENSITY:

4.08	100	4.35	100
4.27	90	3.91	90
4.52	90	4.60	90

The comparison . . . / -

The comparison is not very satisfactory, but indicates that the mineral is almost amorphous and belongs to the Vauxite series.

ORALCOPYALITE:

Only very small amounts of this mineral were available and hence the powder diffraction method was employed, using Fe radiation with the Mn filter and the 57.5 millimetre camera with small ports. The mineral was exposed for 45 minutes to the X-ray beam.

The "d" values obtained are almost identical with the data file figures.

"d" VALUES:

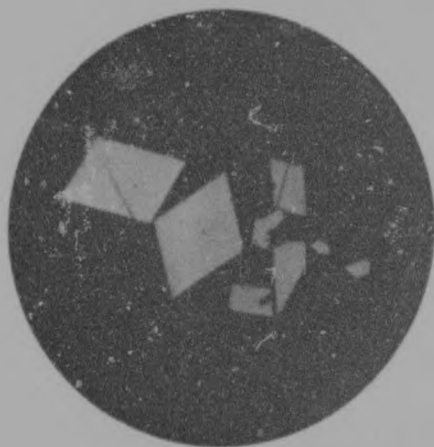
<u>EXPERIMENTAL "d" VALUES:</u>	<u>INTENSITY:</u>	<u>DATA FILE "d" VALUES:</u>	<u>INTENSITY:</u>
3.04	100	3.03	100
1.85	80	1.65	80
1.59	60	1.59	60

PYRRHOTITE:

Pyrrhotite was analysed using Co radiation and Fe filter. The powder diffraction method was employed, with the 57.5 millimetre camera, small ports and 40 minute exposure to radiation. The "d" values obtained for the sample bore no relationship to the figures quoted in the data file. A standard sample was similarly analysed (Pyrrhotite from Sudbury, Canada) . . . / -

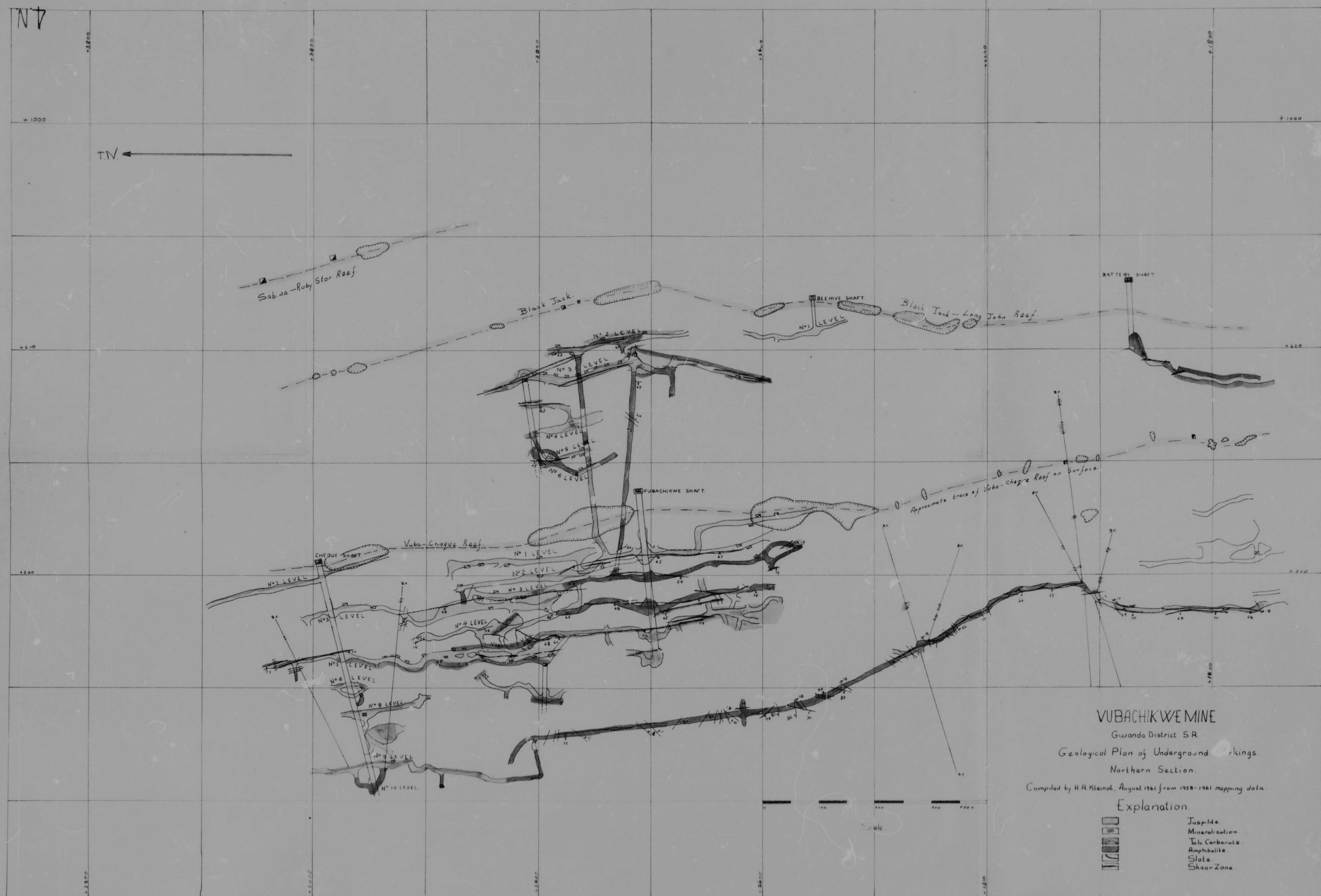


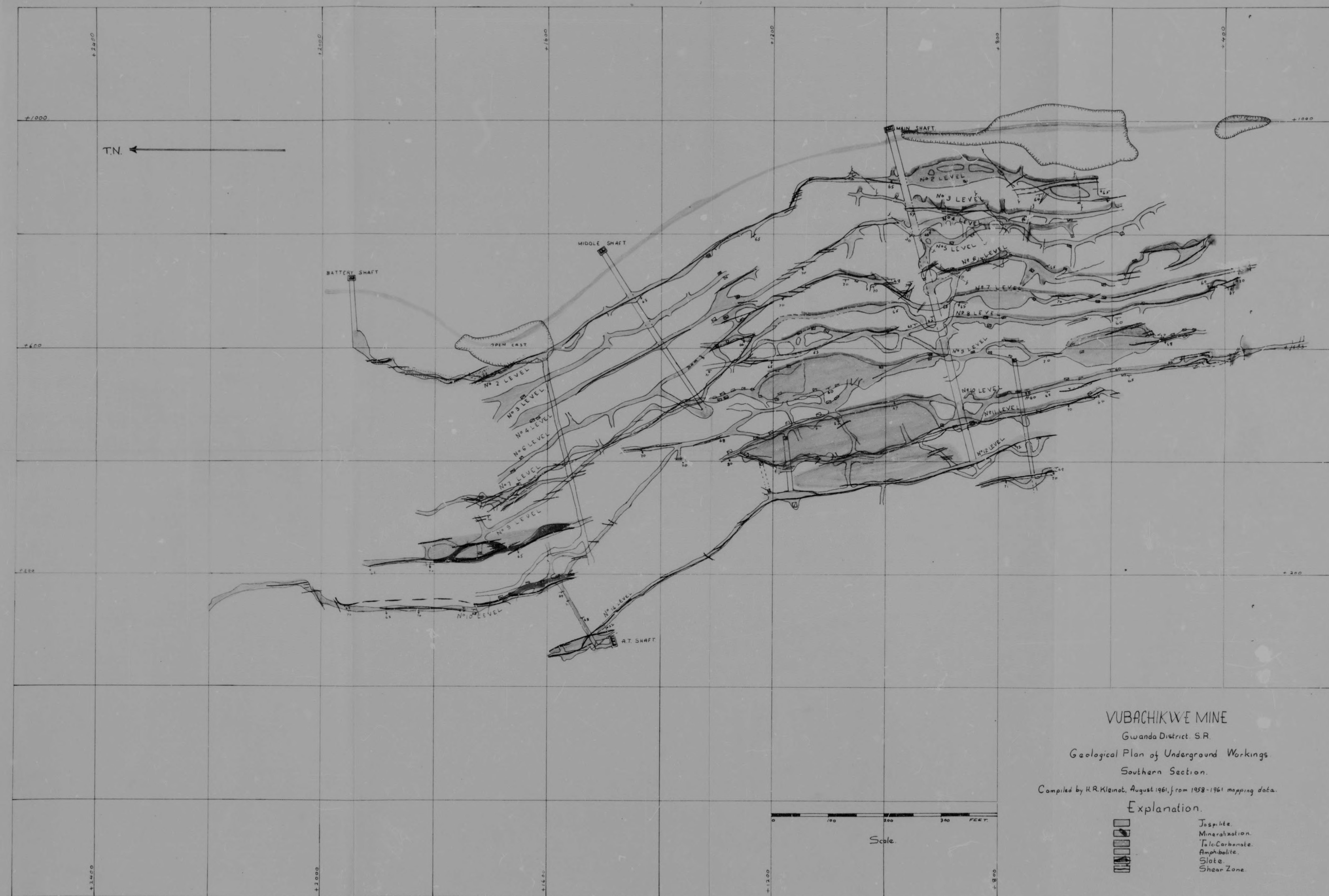
Arsenopyrite partially replaced by quartz
and showing replacement or exsolution
inclusions of pyrrhotite. Polished section.
Plane polarised x 1200.

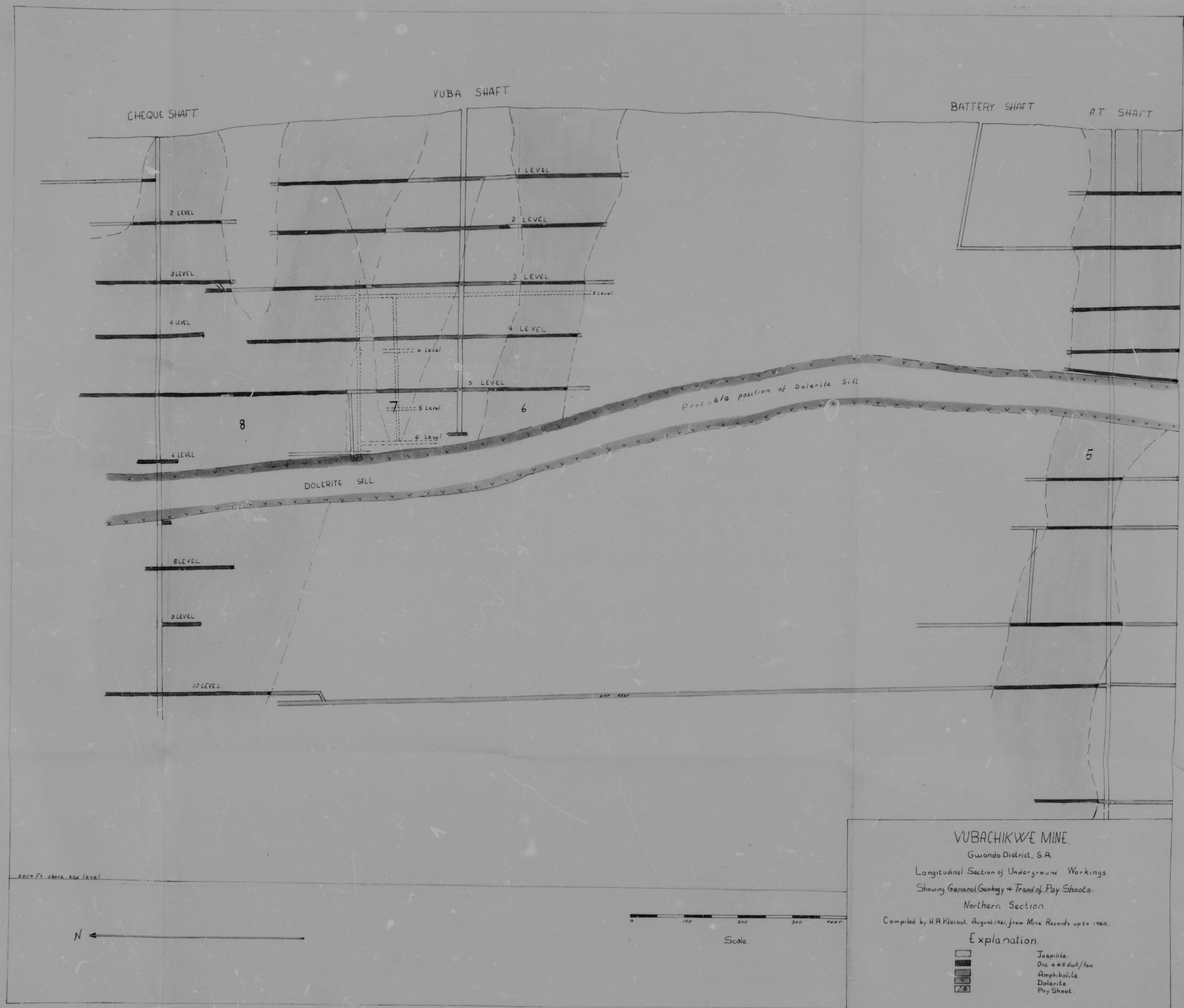


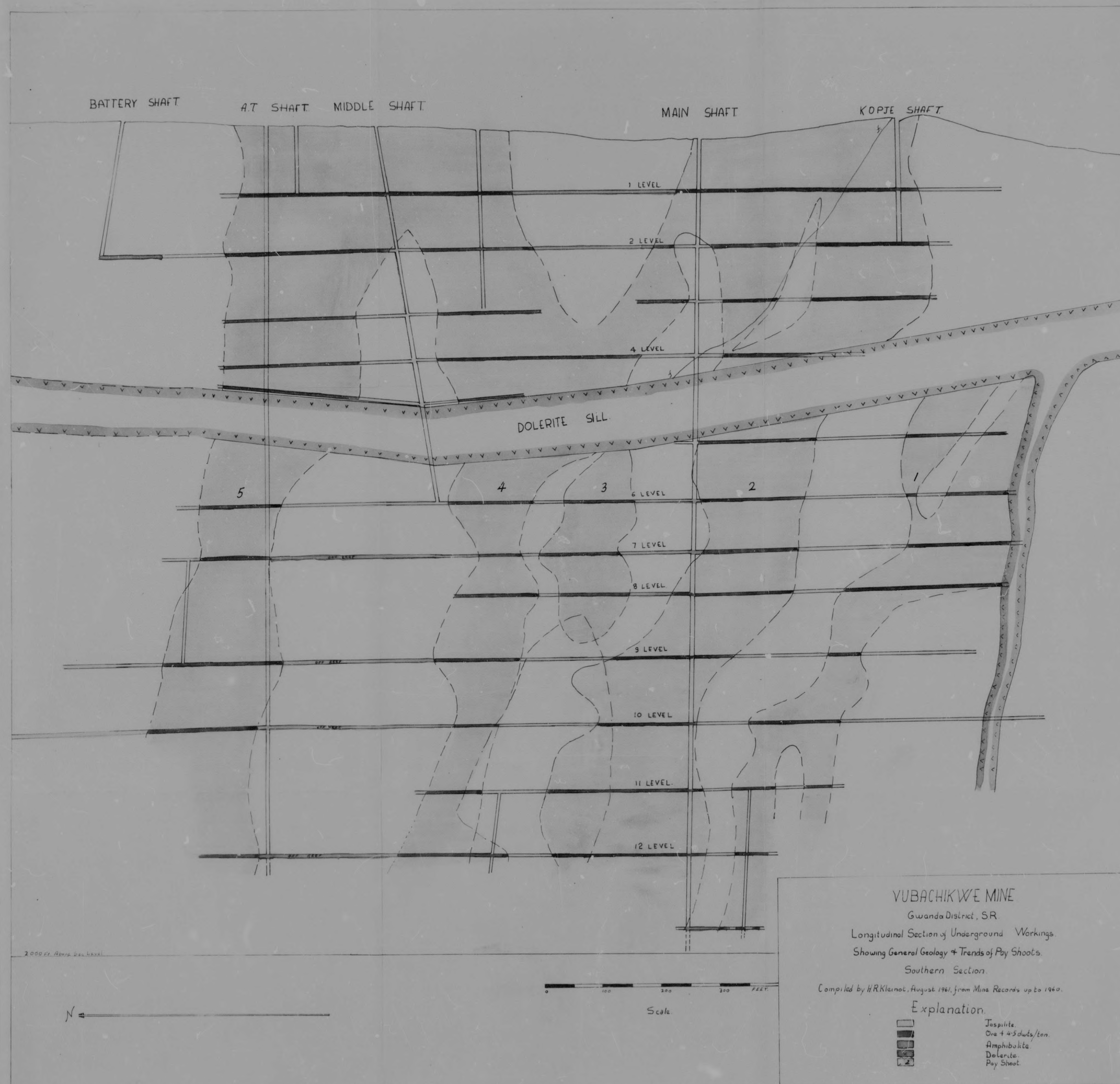
Arsenopyrite in discrete euhedral
crystals in jaspilite ore.
Polished section.
Plane polarised x 250.











VUBACHIKWE MINE

Gwanda District, S.R.

Longitudinal Section of Underground Workings.

Showing General Geology & Trends of Pay Shoots.

Southern Section.

Compiled by H.R. Kleirot, August 1961, from Mine Records up to 1960.

Explanation.

- Jaspilite.
- Orn + 4.5 dwt/ton.
- Amphibolite.
- Dolerite.
- Pay Shoot.

Author Kleinot H R

Name of thesis The structure and mineralization of Vubachikwe Mine - Gwanda, S.R. 1963

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